

2-butyl-5-ethyl-4-methyl-3-thiazoline, cis

Inchi:	InChI=1S/C10H19NS/c1-4-6-7-10-11-8(3)9(5-2)12-10/h9-10H,4-7H2,1-3H3/t9-,10+/m1/s
InchiKey:	KKDDEAQGACNGDN-ZJUUVORDSA-N
Formula:	C10H19NS
SMILES:	CCCCC1N=C(C)C(CC)S1
Mol. weight [g/mol]:	185.33

Physical Properties

Property code	Value	Unit	Source
gf	239.13	kJ/mol	Joback Method
hf	-47.05	kJ/mol	Joback Method
hfus	26.29	kJ/mol	Joback Method
hvap	50.78	kJ/mol	Joback Method
log10ws	-3.67		Crippen Method
logp	3.489		Crippen Method
mcvol	162.930	ml/mol	McGowan Method
pc	2462.92	kPa	Joback Method
rinsol	1383.00		NIST Webbook
tb	544.48	K	Joback Method
tc	759.10	K	Joback Method
tf	377.39	K	Joback Method
vc	0.617	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	402.20	J/mol×K	544.48	Joback Method
cpg	420.89	J/mol×K	580.25	Joback Method
cpg	438.62	J/mol×K	616.02	Joback Method
cpg	455.40	J/mol×K	651.79	Joback Method
cpg	471.25	J/mol×K	687.56	Joback Method
cpg	486.18	J/mol×K	723.33	Joback Method
cpg	500.19	J/mol×K	759.10	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R497539&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvac:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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