

# Phenanthrene, 1,2-dihydro-

<b>Other names:</b>	1,2-Dihydrophenanthrene
<b>Inchi:</b>	InChI=1S/C14H12/c1-3-7-13-11(5-1)9-10-12-6-2-4-8-14(12)13/h1,3-5,7-10H,2,6H2
<b>InchiKey:</b>	DYHJTAONEPZTCO-UHFFFAOYSA-N
<b>Formula:</b>	C14H12
<b>SMILES:</b>	<chem>C1=Cc2c(ccc3ccccc23)CC1</chem>
<b>Mol. weight [g/mol]:</b>	180.25
<b>CAS:</b>	56179-83-0

## Physical Properties

Property code	Value	Unit	Source
gf	353.12	kJ/mol	Joback Method
hf	217.13	kJ/mol	Joback Method
hfus	18.48	kJ/mol	Joback Method
hvap	52.68	kJ/mol	Joback Method
log10ws	-4.80		Crippen Method
logp	3.799		Crippen Method
mcvol	149.740	ml/mol	McGowan Method
pc	3114.04	kPa	Joback Method
rinpol	286.78		NIST Webbook
tb	590.18	K	Joback Method
tc	841.42	K	Joback Method
tf	351.12	K	Joback Method
vc	0.570	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	354.45	J/molxK	590.18	Joback Method
cpg	370.70	J/molxK	632.05	Joback Method
cpg	385.58	J/molxK	673.93	Joback Method
cpg	399.22	J/molxK	715.80	Joback Method
cpg	411.76	J/molxK	757.67	Joback Method
cpg	423.34	J/molxK	799.55	Joback Method
cpg	434.09	J/molxK	841.42	Joback Method

dvisc	0.0016514	Paxs	351.12	Joback Method
dvisc	0.0011961	Paxs	390.96	Joback Method
dvisc	0.0009195	Paxs	430.81	Joback Method
dvisc	0.0007391	Paxs	470.65	Joback Method
dvisc	0.0006147	Paxs	510.49	Joback Method
dvisc	0.0005250	Paxs	550.34	Joback Method
dvisc	0.0004581	Paxs	590.18	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C56179830&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C56179830&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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