

Fumaric acid, nonyl 4-phenylphenyl ester

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|-----------------------------|--|
| Inchi: | InChI=1S/C25H30O4/c1-2-3-4-5-6-7-11-20-28-24(26)18-19-25(27)29-23-16-14-22(15-17 |
| InchiKey: | OVAZBVUBJUFZEK-VHEBQXMUSA-N |
| Formula: | C25H30O4 |
| SMILES: | CCCCCCCCCOC(=O)C=CC(=O)Oc1ccc(-c2ccccc2)cc1 |
| Mol. weight [g/mol]: | 394.50 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -12.81 | kJ/mol | Joback Method |
| hf | -470.12 | kJ/mol | Joback Method |
| hfus | 53.97 | kJ/mol | Joback Method |
| hvap | 94.73 | kJ/mol | Joback Method |
| log10ws | -7.71 | | Crippen Method |
| logp | 6.109 | | Crippen Method |
| mcvol | 326.170 | ml/mol | McGowan Method |
| pc | 1251.26 | kPa | Joback Method |
| rinpol | 3285.00 | | NIST Webbook |
| rinpol | 3285.00 | | NIST Webbook |
| tb | 986.48 | K | Joback Method |
| tc | 1214.03 | K | Joback Method |
| tf | 576.11 | K | Joback Method |
| vc | 1.248 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 1048.20 | J/molxK | 986.48 | Joback Method |
| cpg | 1108.03 | J/molxK | 1176.10 | Joback Method |
| cpg | 1098.30 | J/molxK | 1138.18 | Joback Method |
| cpg | 1087.54 | J/molxK | 1100.25 | Joback Method |
| cpg | 1075.65 | J/molxK | 1062.33 | Joback Method |
| cpg | 1062.56 | J/molxK | 1024.40 | Joback Method |
| cpg | 1116.78 | J/molxK | 1214.03 | Joback Method |
| dvisc | 0.0000241 | Paxs | 986.48 | Joback Method |

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|-------|-----------|------|--------|---------------|
| dvisc | 0.0000311 | Paxs | 918.09 | Joback Method |
| dvisc | 0.0000419 | Paxs | 849.69 | Joback Method |
| dvisc | 0.0000594 | Paxs | 781.30 | Joback Method |
| dvisc | 0.0000901 | Paxs | 712.90 | Joback Method |
| dvisc | 0.0001494 | Paxs | 644.50 | Joback Method |
| dvisc | 0.0002790 | Paxs | 576.11 | Joback Method |

Sources

| | |
|------------------------|---|
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U348215&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |

Legend

| | |
|----------------------------|---|
| cp_g: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log₁₀ws: | Log ₁₀ of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| m_{cvol}: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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