

Phytol propionate

Inchi:	InChI=1S/C23H44O2/c1-7-23(24)25-18-17-22(6)16-10-15-21(5)14-9-13-20(4)12-8-11-19
InchiKey:	KJIHKRXSZYOFGU-IVNWLUGXSA-N
Formula:	C23H44O2
SMILES:	CCC(=O)OCC=C(C)CCCC(C)CCCC(C)CCCC(C)C
Mol. weight [g/mol]:	352.59

Physical Properties

Property code	Value	Unit	Source
gf	-26.79	kJ/mol	Joback Method
hf	-671.26	kJ/mol	Joback Method
hfus	46.44	kJ/mol	Joback Method
hvap	74.82	kJ/mol	Joback Method
log10ws	-7.44		Crippen Method
logp	7.325		Crippen Method
mcvol	338.070	ml/mol	McGowan Method
pc	918.83	kPa	Joback Method
rinqol	2079.00		NIST Webbook
tb	804.65	K	Joback Method
tc	989.69	K	Joback Method
tf	357.09	K	Joback Method
vc	1.310	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1069.77	J/molxK	804.65	Joback Method
cpg	1090.59	J/molxK	835.49	Joback Method
cpg	1110.32	J/molxK	866.33	Joback Method
cpg	1128.99	J/molxK	897.17	Joback Method
cpg	1146.65	J/molxK	928.01	Joback Method
cpg	1163.35	J/molxK	958.85	Joback Method
cpg	1179.12	J/molxK	989.69	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R572259&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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