

# 1,3-Dioxane, 4-(2-methylpropyl)

**Inchi:** InChI=1S/C8H16O2/c1-7(2)5-8-3-4-9-6-10-8/h7-8H,3-6H2,1-2H3  
**InchiKey:** ZRTAVZIDNPIUIG-UHFFFAOYSA-N  
**Formula:** C8H16O2  
**SMILES:** CC(C)CC1CCOCO1  
**Mol. weight [g/mol]:** 144.21

## Physical Properties

Property code	Value	Unit	Source
gf	-133.75	kJ/mol	Joback Method
hf	-423.41	kJ/mol	Joback Method
hfus	20.75	kJ/mol	Joback Method
hvap	42.46	kJ/mol	Joback Method
log10ws	-1.61		Crippen Method
logp	1.796		Crippen Method
mcvol	124.460	ml/mol	McGowan Method
pc	3117.52	kPa	Joback Method
rinpol	938.00		NIST Webbook
rinpol	945.00		NIST Webbook
ripol	1167.00		NIST Webbook
ripol	1181.00		NIST Webbook
ripol	1167.00		NIST Webbook
tb	455.45	K	Joback Method
tc	660.82	K	Joback Method
tf	225.44	K	Joback Method
vc	0.453	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	276.54	J/molxK	455.45	Joback Method
cpg	293.47	J/molxK	489.68	Joback Method
cpg	309.55	J/molxK	523.91	Joback Method
cpg	324.83	J/molxK	558.14	Joback Method
cpg	339.30	J/molxK	592.37	Joback Method

cpg	352.99	J/mol×K	626.60	Joback Method
cpg	365.92	J/mol×K	660.82	Joback Method
dvisc	0.0138481	Paxs	225.44	Joback Method
dvisc	0.0046899	Paxs	263.77	Joback Method
dvisc	0.0020906	Paxs	302.11	Joback Method
dvisc	0.0011179	Paxs	340.44	Joback Method
dvisc	0.0006785	Paxs	378.78	Joback Method
dvisc	0.0004514	Paxs	417.11	Joback Method
dvisc	0.0003217	Paxs	455.45	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R409082&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R409082&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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