

3-(Fluorosulfonyl)benzoic acid

Other names:	Benzoic acid, 3-(fluorosulfonyl)- m-(Fluorosulfonyl)benzoic acid 3-(fluorosulphonyl)benzoic acid
Inchi:	InChI=1S/C7H5FO4S/c8-13(11,12)6-3-1-2-5(4-6)7(9)10/h1-4H,(H,9,10)
InchiKey:	VWYMBWGOJRULOV-UHFFFAOYSA-N
Formula:	C7H5FO4S
SMILES:	O=C(O)c1cccc(S(=O)(=O)F)c1
Mol. weight [g/mol]:	204.18
CAS:	454-95-5

Physical Properties

Property code	Value	Unit	Source
gf	-818.25	kJ/mol	Joback Method
hf	-877.02	kJ/mol	Joback Method
hfus	27.68	kJ/mol	Joback Method
hvap	75.36	kJ/mol	Joback Method
log10ws	-1.65		Crippen Method
logp	1.043		Crippen Method
mcvol	123.030	ml/mol	McGowan Method
pc	5836.07	kPa	Joback Method
tb	584.32	K	Joback Method
tc	778.52	K	Joback Method
tf	357.49	K	Joback Method
vc	0.488	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	275.17	J/mol×K	584.32	Joback Method
cpg	283.53	J/mol×K	616.69	Joback Method
cpg	291.35	J/mol×K	649.05	Joback Method
cpg	298.65	J/mol×K	681.42	Joback Method
cpg	305.42	J/mol×K	713.79	Joback Method
cpg	311.67	J/mol×K	746.15	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C454955&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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