

# 2-Allyl-1,4-dimethoxy-3-methyl-benzene

<b>Inchi:</b>	InChI=1S/C12H16O2/c1-5-6-10-9(2)11(13-3)7-8-12(10)14-4/h5,7-8H,1,6H2,2-4H3
<b>InchiKey:</b>	QVZBRUBDHWOPIM-UHFFFAOYSA-N
<b>Formula:</b>	C12H16O2
<b>SMILES:</b>	C=CCc1c(OC)ccc(OC)c1C
<b>Mol. weight [g/mol]:</b>	192.25

## Physical Properties

Property code	Value	Unit	Source
gf	11.52	kJ/mol	Joback Method
hf	-227.90	kJ/mol	Joback Method
hfus	20.81	kJ/mol	Joback Method
hvap	50.72	kJ/mol	Joback Method
log10ws	-3.26		Crippen Method
logp	2.741		Crippen Method
mcvol	163.620	ml/mol	McGowan Method
pc	2320.31	kPa	Joback Method
rinpol	1601.00		NIST Webbook
tb	557.10	K	Joback Method
tc	760.27	K	Joback Method
tf	331.68	K	Joback Method
vc	0.617	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	378.89	J/molxK	557.10	Joback Method
cpg	393.51	J/molxK	590.96	Joback Method
cpg	407.50	J/molxK	624.82	Joback Method
cpg	420.85	J/molxK	658.69	Joback Method
cpg	433.56	J/molxK	692.55	Joback Method
cpg	445.63	J/molxK	726.41	Joback Method
cpg	457.05	J/molxK	760.27	Joback Method
dvisc	0.0008345	Paxs	331.68	Joback Method
dvisc	0.0005269	Paxs	369.25	Joback Method

dvisc	0.0003622	Paxs	406.82	Joback Method
dvisc	0.0002652	Paxs	444.39	Joback Method
dvisc	0.0002039	Paxs	481.96	Joback Method
dvisc	0.0001628	Paxs	519.53	Joback Method
dvisc	0.0001340	Paxs	557.10	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U187537&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U187537&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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