

# Octadecane, 9-ethyl-9-heptyl-

<b>Other names:</b>	9-Ethyl-9-n-heptyloctadecane
<b>Inchi:</b>	InChI=1S/C27H56/c1-5-9-12-15-17-20-23-26-27(8-4,24-21-18-14-11-7-3)25-22-19-16-13
<b>InchiKey:</b>	LRJHOBBNMAGOIG-UHFFFAOYSA-N
<b>Formula:</b>	C27H56
<b>SMILES:</b>	CCCCCCCCC(CC)(CCCCCCC)CCCCCCCC
<b>Mol. weight [g/mol]:</b>	380.73
<b>CAS:</b>	55282-27-4

## Physical Properties

Property code	Value	Unit	Source
gf	179.30	kJ/mol	Joback Method
hf	-609.36	kJ/mol	Joback Method
hfus	58.27	kJ/mol	Joback Method
hvap	74.40	kJ/mol	Joback Method
log10ws	-10.88		Crippen Method
logp	10.635		Crippen Method
mcvol	391.290	ml/mol	McGowan Method
pc	695.81	kPa	Joback Method
tb	813.93	K	Joback Method
tc	996.49	K	Joback Method
tf	193.00 ± 5.00	K	NIST Webbook
vc	1.536	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1282.03	J/mol×K	813.93	Joback Method
cpg	1306.42	J/mol×K	844.36	Joback Method
cpg	1329.60	J/mol×K	874.78	Joback Method
cpg	1351.65	J/mol×K	905.21	Joback Method
cpg	1372.63	J/mol×K	935.64	Joback Method
cpg	1392.59	J/mol×K	966.07	Joback Method
cpg	1411.62	J/mol×K	996.49	Joback Method
dvisc	0.0016537	Paxs	396.47	Joback Method

dvisc	0.0005159	Paxs	466.05	Joback Method
dvisc	0.0002178	Paxs	535.62	Joback Method
dvisc	0.0001122	Paxs	605.20	Joback Method
dvisc	0.0000662	Paxs	674.78	Joback Method
dvisc	0.0000431	Paxs	744.35	Joback Method
dvisc	0.0000302	Paxs	813.93	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C55282274&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C55282274&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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