

# 1,3-Pentadiene

<b>Other names:</b>	Penta-1,3-diene Piperylene 1-Methylbutadiene CH <sub>2</sub> =CHCH=CHCH <sub>3</sub> Rcra waste number U186
<b>Inchi:</b>	InChI=1S/C5H8/c1-3-5-4-2/h3-5H,1H2,2H3
<b>InchiKey:</b>	PMJHHCWVYXUKFD-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>5</sub> H <sub>8</sub>
<b>SMILES:</b>	C=CC=CC
<b>Mol. weight [g/mol]:</b>	68.12
<b>CAS:</b>	504-60-9

## Physical Properties

Property code	Value	Unit	Source
gf	159.28	kJ/mol	Joback Method
hf	96.12	kJ/mol	Joback Method
hfus	7.63	kJ/mol	Joback Method
hvap	26.01	kJ/mol	Joback Method
ie	8.60	eV	NIST Webbook
log10ws	-1.62		Crippen Method
logp	1.748		Crippen Method
mcvol	72.710	ml/mol	McGowan Method
pc	3886.79	kPa	Joback Method
rinpol	525.00		NIST Webbook
rinpol	526.00		NIST Webbook
rinpol	515.00		NIST Webbook
rinpol	516.00		NIST Webbook
rinpol	515.00		NIST Webbook
rinpol	515.00		NIST Webbook
rinpol	542.00		NIST Webbook
rinpol	515.00		NIST Webbook
rinpol	526.00		NIST Webbook
rinpol	521.00		NIST Webbook
ripol	689.00		NIST Webbook
ripol	689.00		NIST Webbook
tb	314.00 ± 2.00	K	NIST Webbook
tb	315.30 ± 2.00	K	NIST Webbook

tb	315.00 ± 3.00	K	NIST Webbook
tb	315.60 ± 2.00	K	NIST Webbook
tb	315.00 ± 3.00	K	NIST Webbook
tb	315.70 ± 2.00	K	NIST Webbook
tb	314.40 ± 2.00	K	NIST Webbook
tb	316.00 ± 4.00	K	NIST Webbook
tb	314.90 ± 1.50	K	NIST Webbook
tb	315.70 ± 2.00	K	NIST Webbook
tb	315.00	K	NIST Webbook
tb	315.20	K	NIST Webbook
tc	490.29	K	Joback Method
tf	139.27	K	Joback Method
vc	0.277	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	100.21	J/mol×K	314.64	Joback Method
cpg	108.47	J/mol×K	343.91	Joback Method
cpg	116.31	J/mol×K	373.19	Joback Method
cpg	123.76	J/mol×K	402.46	Joback Method
cpg	130.84	J/mol×K	431.74	Joback Method
cpg	137.57	J/mol×K	461.01	Joback Method
cpg	143.95	J/mol×K	490.29	Joback Method
dvisc	0.0024945	Paxs	139.27	Joback Method
dvisc	0.0010531	Paxs	168.50	Joback Method
dvisc	0.0005737	Paxs	197.73	Joback Method
dvisc	0.0003655	Paxs	226.95	Joback Method
dvisc	0.0002581	Paxs	256.18	Joback Method
dvisc	0.0001957	Paxs	285.41	Joback Method
dvisc	0.0001562	Paxs	314.64	Joback Method

## Sources

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C504609&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/46-917-3/1-3-Pentadiene.pdf>

Generated by Cheméo on 2024-05-22 01:33:19.581914987 +0000 UTC m=+18630848.502492308.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.