

# Metobromuron

<b>Other names:</b>	Urea, N'-(4-bromophenyl)-N-methoxy-N-methyl- Urea, 3-(p-bromophenyl)-1-methoxy-1-methyl- C 3126 Metbromuron Patoran 3-(p-Bromophenyl)-1-Methoxy-1-methylurea 3-(4-Bromophenyl)-1-methoxy-1-methylurea 3-(4-Bromophenyl)-1-methoxyharnstoff N'-(4-Bromophenyl)-N-methoxy-N-methylurea 3-(p-Bromophenyl)-1-methyl-1-methoxyurea CIBA-3126 Pattonex Patoran FL Monobromuron N-(p-Bromophenyl)-N'-methyl-N'-methoxyurea
<b>Inchi:</b>	InChI=1S/C9H11BrN2O2/c1-12(14-2)9(13)11-8-5-3-7(10)4-6-8/h3-6H,1-2H3,(H,11,13)
<b>InchiKey:</b>	WLFDQEVORAMCIM-UHFFFAOYSA-N
<b>Formula:</b>	C9H11BrN2O
<b>SMILES:</b>	CON(C)C(=O)Nc1ccc(Br)cc1
<b>Mol. weight [g/mol]:</b>	243.10
<b>CAS:</b>	3060-89-7

## Physical Properties

Property code	Value	Unit	Source
gf	108.25	kJ/mol	Joback Method
hf	-101.50	kJ/mol	Joback Method
hfus	28.91	kJ/mol	Joback Method
hvap	62.64	kJ/mol	Joback Method
log10ws	-2.88		Crippen Method
logp	2.474		Crippen Method
mcvol	158.810	ml/mol	McGowan Method
pc	3718.02	kPa	Joback Method
tb	642.04	K	Joback Method
tc	868.44	K	Joback Method
tf	368.88 ± 0.20	K	NIST Webbook
tf	369.00 ± 0.20	K	NIST Webbook
vc	0.571	m3/kmol	Joback Method

---

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	414.29	J/mol×K	792.98	Joback Method
cpg	422.94	J/mol×K	830.71	Joback Method
cpg	371.94	J/mol×K	642.04	Joback Method
cpg	383.77	J/mol×K	679.77	Joback Method
cpg	394.74	J/mol×K	717.51	Joback Method
cpg	404.91	J/mol×K	755.24	Joback Method
cpg	430.88	J/mol×K	868.44	Joback Method
hfust	24.44	kJ/mol	368.30	NIST Webbook
hfust	24.44	kJ/mol	368.30	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3060897&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3060897&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/46-924-5/Metobromuron.pdf>

Generated by Cheméo on 2024-04-29 14:51:01.677005832 +0000 UTC m=+16691510.597583154.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.