

# Cubane-1,4-dicarboxylic acid dimethyl ester

<b>Inchi:</b>	InChI=1S/C12H12O4/c1-15-9(13)11-3-6-4(11)8-5(11)7(3)12(6,8)10(14)16-2/h3-8H,1-2H3
<b>InchiKey:</b>	OXBMFCGRQJNAOY-UHFFFAOYSA-N
<b>Formula:</b>	C12H12O4
<b>SMILES:</b>	COC(=O)C12C3C4C1C1C2C3C41C(=O)OC
<b>Mol. weight [g/mol]:</b>	220.22
<b>CAS:</b>	29412-62-2

## Physical Properties

Property code	Value	Unit	Source
chs	-6218.10 ± 1.40	kJ/mol	NIST Webbook
gf	-63.34	kJ/mol	Joback Method
hf	-404.03	kJ/mol	Joback Method
hfs	-219.00 ± 2.10	kJ/mol	NIST Webbook
hfus	28.40	kJ/mol	Joback Method
hsub	118.90 ± 3.80	kJ/mol	NIST Webbook
hsub	117.20 ± 3.90	kJ/mol	NIST Webbook
hvap	88.50 ± 2.20	kJ/mol	NIST Webbook
log10ws	-0.11		Crippen Method
logp	0.070		Crippen Method
mcvol	140.520	ml/mol	McGowan Method
pc	3045.68	kPa	Joback Method
tb	616.82	K	Joback Method
tc	825.08	K	Joback Method
tf	437.80 ± 0.50	K	NIST Webbook
vc	0.590	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	444.48	J/mol×K	651.53	Joback Method
cpg	455.67	J/mol×K	686.24	Joback Method
cpg	466.42	J/mol×K	720.95	Joback Method
cpg	477.05	J/mol×K	755.66	Joback Method
cpg	487.85	J/mol×K	790.37	Joback Method

cpg	432.55	J/mol×K	616.82	Joback Method
cpg	499.14	J/mol×K	825.08	Joback Method
cps	251.10	J/mol×K	298.15	NIST Webbook
hfust	41.00	kJ/mol	437.80	NIST Webbook
hfust	38.10	kJ/mol	438.20	NIST Webbook
hfust	41.00	kJ/mol	437.80	NIST Webbook
sfust	93.70	J/mol×K	437.80	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C29412622&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C29412622&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cps:</b>	Solid phase heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>sfust:</b>	Entropy of fusion at a given temperature
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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