

# Benzenepropanamine

|                             |   |
|-----------------------------|---|
| <b>Other names:</b>         | 1-Amino-3-phenylpropane<br>3-Phenyl-1-aminopropane<br>3-Phenyl-1-propanamine<br>3-Phenyl-1-propylamine<br>3-Phenyl-n-propylamine<br>3-Phenylpropanamine<br>3-Phenylpropylamine<br>Hydrocinnamylamine<br>NSC 87080<br>Propylamine, 3-phenyl-<br>«gamma»-Phenyl-n-propylamine<br>«gamma»-Phenylpropylamine<br>Â«gammaÂ»-Phenyl-n-propylamine<br>Â«gammaÂ»-Phenylpropylamine |
| <b>Inchi:</b>               | InChI=1S/C9H13N/c10-8-4-7-9-5-2-1-3-6-9/h1-3,5-6H,4,7-8,10H2  |
| <b>InchiKey:</b>            | LYUQWQRDLDVQGA-UHFFFAOYSA-N   |
| <b>Formula:</b>             | C9H13N  |
| <b>SMILES:</b>              | NCCCC1CCCC1   |
| <b>Mol. weight [g/mol]:</b> | 135.21  |
| <b>CAS:</b>                 | 2038-57-5   |

## Physical Properties

| Property code | Value         | Unit   | Source         |
|---------------|---------------|--------|----------------|
| basg          | 920.00 ± 8.00 | kJ/mol | NIST Webbook   |
| gf            | 203.76        | kJ/mol | Joback Method  |
| hf            | 41.23         | kJ/mol | Joback Method  |
| hfus          | 18.30         | kJ/mol | Joback Method  |
| hvap          | 48.55         | kJ/mol | Joback Method  |
| ie            | 8.89 ± 0.12   | eV     | NIST Webbook   |
| log10ws       | -2.13         |        | Crippen Method |
| logp          | 1.578         |        | Crippen Method |
| mcvol         | 123.890       | ml/mol | McGowan Method |
| pc            | 3484.76       | kPa    | Joback Method  |
| rinpol        | 1205.00       |        | NIST Webbook   |
| rinpol        | 1205.00       |        | NIST Webbook   |
| tb            | 494.20        | K      | NIST Webbook   |
| tc            | 723.91        | K      | Joback Method  |

|    |        |                      |               |
|----|--------|----------------------|---------------|
| tf | 300.87 | K                    | Joback Method |
| vc | 0.461  | m <sup>3</sup> /kmol | Joback Method |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 333.59 | J/mol×K | 687.35          | Joback Method |
| cpg           | 271.45 | J/mol×K | 504.53          | Joback Method |
| cpg           | 285.54 | J/mol×K | 541.09          | Joback Method |
| cpg           | 298.76 | J/mol×K | 577.66          | Joback Method |
| cpg           | 311.15 | J/mol×K | 614.22          | Joback Method |
| cpg           | 322.74 | J/mol×K | 650.78          | Joback Method |
| cpg           | 343.71 | J/mol×K | 723.91          | Joback Method |
| cpl           | 265.59 | J/mol×K | 298.15          | NIST Webbook  |

## Correlations

| Information                 | Value                         |
|-----------------------------|-------------------------------|
| Property code               | pvap                          |
| Equation                    | $\ln(P_{vp}) = A + B/(T + C)$ |
| Coeff. A                    | 1.46446e+01                   |
| Coeff. B                    | -4.16770e+03                  |
| Coeff. C                    | -7.85240e+01                  |
| Temperature range (K), min. | 368.81                        |
| Temperature range (K), max. | 525.07                        |

## Sources

|   |   |
|---|---|
| <b>McGowan Method:</b>                      | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>   |
| <b>NIST Webbook:</b>                        | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2038575&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2038575&amp;Units=SI</a>   |
| <b>The Yaws Handbook of Vapor Pressure:</b> | <a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a> |
| <b>Crippen Method:</b>                      | <a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>   |
| <b>Crippen Method:</b>                      | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>   |
| <b>Joback Method:</b>                       | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>   |

# Legend

|                 |   |
|-----------------|---|
| <b>basg:</b>    | Gas basicity                                    |
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>cpl:</b>     | Liquid phase heat capacity                      |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>ie:</b>      | Ionization energy                               |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>pvap:</b>    | Vapor pressure                                  |
| <b>rinpola:</b> | Non-polar retention indices                     |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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