

# Benzaldehyde, 4-fluoro-

<b>Other names:</b>	Benzaldehyde, p-fluoro- p-Fluorobenzaldehyde 4-Fluorobenzaldehyde para-Fluorobenzaldehyde
<b>Inchi:</b>	InChI=1S/C7H5FO/c8-7-3-1-6(5-9)2-4-7/h1-5H
<b>InchiKey:</b>	UOQXIWFQBQSVDP-UFHFFFAOYSA-N
<b>Formula:</b>	C7H5FO
<b>SMILES:</b>	O=Cc1ccc(F)cc1
<b>Mol. weight [g/mol]:</b>	124.11
<b>CAS:</b>	459-57-4

## Physical Properties

Property code	Value	Unit	Source
affp	827.10	kJ/mol	NIST Webbook
basg	795.30	kJ/mol	NIST Webbook
ea	0.49 ± 0.02	eV	NIST Webbook
gf	-183.49	kJ/mol	Joback Method
hf	-244.44	kJ/mol	Joback Method
hfus	12.91	kJ/mol	Joback Method
h vap	40.02	kJ/mol	Joback Method
ie	9.60	eV	NIST Webbook
log10ws	-2.04		Crippen Method
logp	1.638		Crippen Method
m cvol	89.070	ml/mol	McGowan Method
pc	4189.32	kPa	Joback Method
rinpol	942.50		NIST Webbook
rinpol	948.30		NIST Webbook
rinpol	936.00		NIST Webbook
tb	452.50 ± 0.50	K	NIST Webbook
tc	647.38	K	Joback Method
tf	250.18	K	Joback Method
vc	0.354	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	164.02	J/mol×K	439.15	Joback Method
cpg	173.14	J/mol×K	473.86	Joback Method
cpg	181.71	J/mol×K	508.56	Joback Method
cpg	189.77	J/mol×K	543.27	Joback Method
cpg	197.34	J/mol×K	577.97	Joback Method
cpg	204.42	J/mol×K	612.68	Joback Method
cpg	211.05	J/mol×K	647.38	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	454.20	K	101.00	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C459574&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C459574&amp;Units=SI</a>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>cpg:</b>	Ideal gas heat capacity
<b>ea:</b>	Electron affinity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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