

# (Z)-4,4-Dimethylhex-2-ene

<b>Other names:</b>	2-Hexene, 4,4-dimethyl, cis
<b>Inchi:</b>	InChI=1S/C8H16/c1-5-7-8(3,4)6-2/h5,7H,6H2,1-4H3/b7-5-
<b>InchiKey:</b>	OQEVAISXHCRRQGF-ALCCZGGFSA-N
<b>Formula:</b>	C8H16
<b>SMILES:</b>	CC=CC(C)(C)CC
<b>Mol. weight [g/mol]:</b>	112.21

## Physical Properties

Property code	Value	Unit	Source
gf	99.54	kJ/mol	Joback Method
hf	-99.98	kJ/mol	Joback Method
hfus	9.26	kJ/mol	Joback Method
hvap	32.06	kJ/mol	Joback Method
log10ws	-2.78		Crippen Method
logp	2.999		Crippen Method
mcpvol	119.280	ml/mol	McGowan Method
pc	2726.86	kPa	Joback Method
rinpol	745.60		NIST Webbook
rinpol	744.00		NIST Webbook
rinpol	753.50		NIST Webbook
tb	383.37	K	Joback Method
tc	566.77	K	Joback Method
tf	177.26	K	Joback Method
vc	0.453	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	219.13	J/molxK	383.37	Joback Method
cpg	233.64	J/molxK	413.94	Joback Method
cpg	247.38	J/molxK	444.50	Joback Method
cpg	260.39	J/molxK	475.07	Joback Method
cpg	272.68	J/molxK	505.64	Joback Method
cpg	284.31	J/molxK	536.21	Joback Method

cpg	295.30	J/mol×K	566.77	Joback Method
dvisc	0.0110015	Paxs	177.26	Joback Method
dvisc	0.0034054	Paxs	211.61	Joback Method
dvisc	0.0014626	Paxs	245.96	Joback Method
dvisc	0.0007728	Paxs	280.31	Joback Method
dvisc	0.0004693	Paxs	314.67	Joback Method
dvisc	0.0003144	Paxs	349.02	Joback Method
dvisc	0.0002263	Paxs	383.37	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R293164&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R293164&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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