

# Butanoic acid, 3-chloro, 2-methylpropyl ester

<b>Other names:</b>	Isobutyl 3-chlorobutyrate
<b>Inchi:</b>	InChI=1S/C8H15ClO2/c1-6(2)5-11-8(10)4-7(3)9/h6-7H,4-5H2,1-3H3
<b>InchiKey:</b>	HRVVBUFBYXPDEB-UHFFFAOYSA-N
<b>Formula:</b>	C8H15ClO2
<b>SMILES:</b>	CC(C)COC(=O)CC(C)Cl
<b>Mol. weight [g/mol]:</b>	178.66
<b>CAS:</b>	62108-76-3

## Physical Properties

Property code	Value	Unit	Source
chl	-4691.90 ± 8.40	kJ/mol	NIST Webbook
gf	-234.25	kJ/mol	Joback Method
hf	-571.10 ± 9.60	kJ/mol	NIST Webbook
hfl	-623.40 ± 8.40	kJ/mol	NIST Webbook
hfus	16.41	kJ/mol	Joback Method
hvap	52.30 ± 4.20	kJ/mol	NIST Webbook
log10ws	-2.06		Crippen Method
logp	2.203		Crippen Method
mcvol	143.260	ml/mol	McGowan Method
pc	2600.43	kPa	Joback Method
rinpol	1105.00		NIST Webbook
rinpol	1097.00		NIST Webbook
rinpol	1092.00		NIST Webbook
rinpol	1081.00		NIST Webbook
rinpol	1085.00		NIST Webbook
ripol	1468.00		NIST Webbook
ripol	1488.00		NIST Webbook
ripol	1464.00		NIST Webbook
ripol	1472.00		NIST Webbook
ripol	1472.00		NIST Webbook
tb	495.28	K	Joback Method
tc	683.89	K	Joback Method
tf	252.00	K	Joback Method
vc	0.544	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	308.11	J/molxK	495.28	Joback Method
cpg	320.63	J/molxK	526.72	Joback Method
cpg	332.62	J/molxK	558.15	Joback Method
cpg	344.09	J/molxK	589.59	Joback Method
cpg	355.06	J/molxK	621.02	Joback Method
cpg	365.51	J/molxK	652.46	Joback Method
cpg	375.47	J/molxK	683.89	Joback Method
dvisc	0.0061890	Paxs	252.00	Joback Method
dvisc	0.0024580	Paxs	292.55	Joback Method
dvisc	0.0012223	Paxs	333.09	Joback Method
dvisc	0.0007073	Paxs	373.64	Joback Method
dvisc	0.0004556	Paxs	414.19	Joback Method
dvisc	0.0003174	Paxs	454.73	Joback Method
dvisc	0.0002346	Paxs	495.28	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C62108763&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C62108763&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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