

# 2,2',2'',2'''-[Ethane-1,2-diylbis(azanetriyl)]tetraacetate

Other names:

Tetramethyl 2,2',2'',2'''-[ethane-1,2-diylbis(azanetriyl)]tetraacetate

Inchi: InChI=1S/C14H24N2O8/c1-21-11(17)7-15(8-12(18)22-2)5-6-16(9-13(19)23-3)10-14(20)2

InchiKey: LZNZCMHEVACKGP-UHFFFAOYSA-N

Formula: C14H24N2O8

SMILES: COC(=O)CN(CCN(CC(=O)OC)CC(=O)OC)CC(=O)OC

Mol. weight [g/mol]: 348.35

## Physical Properties

Property code	Value	Unit	Source
gf	-647.12	kJ/mol	Joback Method
hf	-1176.43	kJ/mol	Joback Method
hfus	49.21	kJ/mol	Joback Method
hvap	87.47	kJ/mol	Joback Method
log10ws	1.73		Crippen Method
logp	-1.718		Crippen Method
mcvol	257.840	ml/mol	McGowan Method
pc	1762.45	kPa	Joback Method
rinpol	2158.00		NIST Webbook
tb	849.76	K	Joback Method
tc	1044.90	K	Joback Method
tf	601.12	K	Joback Method
vc	0.952	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	798.08	J/molxK	849.76	Joback Method
cpg	810.79	J/molxK	882.28	Joback Method
cpg	822.43	J/molxK	914.81	Joback Method
cpg	832.97	J/molxK	947.33	Joback Method
cpg	842.43	J/molxK	979.86	Joback Method
cpg	850.79	J/molxK	1012.38	Joback Method
cpg	858.05	J/molxK	1044.90	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U378707&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U378707&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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