

# Ethanedial, dioxime

<b>Other names:</b>	Glyoxal, dioxime Ethanedione dioxime Glyoxime Pik-off Cga-22911
<b>Inchi:</b>	InChI=1S/C2H4N2O2/c5-3-1-2-4-6/h1-2,5-6H
<b>InchiKey:</b>	LJHFIVQEAFURQ-UHFFFAOYSA-N
<b>Formula:</b>	C2H4N2O2
<b>SMILES:</b>	ON=CC=NO
<b>Mol. weight [g/mol]:</b>	88.07
<b>CAS:</b>	557-30-2

## Physical Properties

Property code	Value	Unit	Source
chs	-1272.00	kJ/mol	NIST Webbook
hf	-224.63	kJ/mol	Joback Method
hvap	60.03	kJ/mol	Joback Method
log10ws	1.70		Crippen Method
logp	-0.094		Crippen Method
mcvol	62.140	ml/mol	McGowan Method
pc	4849.43	kPa	Joback Method
tb	582.88	K	Joback Method
tc	776.66	K	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C557302&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C557302&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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