

Phenethylamine, 2,5-dimethoxy-4-methylthio, N-acetyl

Inchi:

InchiKey:

Formula:

SMILES:

Mol. weight [g/mol]:

InChI=1S/C13H19NO3S/c1-9(15)14-6-5-10-7-12(17-3)13(18-4)8-11(10)16-2/h7-8H,5-6H

IXIKVQQQZVKSLD-UHFFFAOYSA-N

C13H19NO3S

COc1cc(SC)c(OC)cc1CCNC(C)=O

269.36

Physical Properties

Property code	Value	Unit	Source
gf	-74.31	kJ/mol	Joback Method
hf	-391.21	kJ/mol	Joback Method
hfus	35.50	kJ/mol	Joback Method
hvap	73.61	kJ/mol	Joback Method
log10ws	-3.06		Crippen Method
logp	2.104		Crippen Method
mcvol	209.910	ml/mol	McGowan Method
pc	2222.89	kPa	Joback Method
rinpol	2230.00		NIST Webbook
rinpol	2230.00		NIST Webbook
tb	756.12	K	Joback Method
tc	974.23	K	Joback Method
tf	481.70	K	Joback Method
vc	0.786	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	576.89	J/mol×K	756.12	Joback Method
cpg	590.98	J/mol×K	792.47	Joback Method
cpg	604.07	J/mol×K	828.82	Joback Method
cpg	616.15	J/mol×K	865.17	Joback Method
cpg	627.19	J/mol×K	901.52	Joback Method
cpg	637.20	J/mol×K	937.88	Joback Method
cpg	646.16	J/mol×K	974.23	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R418541&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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