

# 1-(2-Methylphenyl)ethanol

<b>Other names:</b>	o-Tolyl methylcarbinol Methyl o-tolyl carbinol Benzenemethanol, «alpha»,2-dimethyl- «alpha»,2-Dimethylbenzyl alcohol
<b>Inchi:</b>	InChI=1S/C9H12O/c1-7-5-3-4-6-9(7)8(2)10/h3-6,8,10H,1-2H3
<b>InchiKey:</b>	SDCBYRLJYGORNK-UHFFFAOYSA-N
<b>Formula:</b>	C9H12O
<b>SMILES:</b>	Cc1ccccc1C(C)O
<b>Mol. weight [g/mol]:</b>	136.19
<b>CAS:</b>	7287-82-3

## Physical Properties

Property code	Value	Unit	Source
gf	-11.58	kJ/mol	Joback Method
hf	-161.54	kJ/mol	Joback Method
hfus	13.28	kJ/mol	Joback Method
hvap	54.86	kJ/mol	Joback Method
log10ws	-2.47		Crippen Method
logp	2.048		Crippen Method
mcvol	119.780	ml/mol	McGowan Method
pc	3602.88	kPa	Joback Method
tb	528.72	K	Joback Method
tc	728.91	K	Joback Method
tf	275.95	K	Joback Method
vc	0.445	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	266.77	J/molxK	528.72	Joback Method
cpg	278.42	J/molxK	562.08	Joback Method
cpg	289.43	J/molxK	595.45	Joback Method
cpg	299.83	J/molxK	628.81	Joback Method
cpg	309.65	J/molxK	662.18	Joback Method

cpg	318.90	J/mol×K	695.54	Joback Method
cpg	327.61	J/mol×K	728.91	Joback Method
dvisc	0.0169073	Paxs	275.95	Joback Method
dvisc	0.0042083	Paxs	318.08	Joback Method
dvisc	0.0014502	Paxs	360.21	Joback Method
dvisc	0.0006246	Paxs	402.34	Joback Method
dvisc	0.0003156	Paxs	444.46	Joback Method
dvisc	0.0001795	Paxs	486.59	Joback Method
dvisc	0.0001117	Paxs	528.72	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7287823&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7287823&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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