

# Furan-2-carboxaldehyde, 3,5-dimethyl

<b>Inchi:</b>	InChI=1S/C8H10O/c1-6-3-7(2)8(4-6)5-9/h3,5H,4H2,1-2H3
<b>InchiKey:</b>	MWKCFGKWTQJWNR-UHFFFAOYSA-N
<b>Formula:</b>	C8H10O
<b>SMILES:</b>	CC1=CC(C)=C(C=O)C1
<b>Mol. weight [g/mol]:</b>	122.16

## Physical Properties

Property code	Value	Unit	Source
gf	-7.75	kJ/mol	Joback Method
hf	-132.06	kJ/mol	Joback Method
hfus	12.91	kJ/mol	Joback Method
hvap	43.26	kJ/mol	Joback Method
log10ws	-2.05		Crippen Method
logp	1.852		Crippen Method
mcvol	105.690	ml/mol	McGowan Method
pc	3577.07	kPa	Joback Method
rinpol	1021.00		NIST Webbook
rinpol	1021.00		NIST Webbook
rinpol	1019.00		NIST Webbook
rinpol	1019.00		NIST Webbook
ripol	1660.00		NIST Webbook
ripol	1664.00		NIST Webbook
tb	464.31	K	Joback Method
tc	672.36	K	Joback Method
tf	276.14	K	Joback Method
vc	0.414	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	211.03	J/molxK	464.31	Joback Method
cpg	222.01	J/molxK	498.98	Joback Method
cpg	232.42	J/molxK	533.66	Joback Method
cpg	242.29	J/molxK	568.33	Joback Method

cpg	251.63	J/mol×K	603.01	Joback Method
cpg	260.46	J/mol×K	637.68	Joback Method
cpg	268.81	J/mol×K	672.36	Joback Method
dvisc	0.0015995	Paxs	276.14	Joback Method
dvisc	0.0010775	Paxs	307.50	Joback Method
dvisc	0.0007809	Paxs	338.86	Joback Method
dvisc	0.0005977	Paxs	370.22	Joback Method
dvisc	0.0004770	Paxs	401.59	Joback Method
dvisc	0.0003933	Paxs	432.95	Joback Method
dvisc	0.0003328	Paxs	464.31	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R68263&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R68263&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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