

# Hexanoic acid, 3,5,5-trimethyl-

<b>Other names:</b>	3,5,5-Trimethylhexanoic acid
<b>Inchi:</b>	InChI=1S/C9H18O2/c1-7(5-8(10)11)6-9(2,3)4/h7H,5-6H2,1-4H3,(H,10,11)
<b>InchiKey:</b>	OILUAKBAMVLXGF-UHFFFAOYSA-N
<b>Formula:</b>	C9H18O2
<b>SMILES:</b>	CC(CC(=O)O)CC(C)(C)C
<b>Mol. weight [g/mol]:</b>	158.24
<b>CAS:</b>	3302-10-1

## Physical Properties

Property code	Value	Unit	Source
gf	-240.44	kJ/mol	Joback Method
hf	-507.93	kJ/mol	Joback Method
hfus	13.82	kJ/mol	Joback Method
hvap	57.37	kJ/mol	Joback Method
log10ws	-2.21		Crippen Method
logp	2.533		Crippen Method
mcvol	145.110	ml/mol	McGowan Method
pc	2770.08	kPa	Joback Method
rinpol	1139.00		NIST Webbook
tb	547.70	K	Joback Method
tc	727.60	K	Joback Method
tf	289.36	K	Joback Method
vc	0.547	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	358.91	J/molxK	547.70	Joback Method
cpg	371.56	J/molxK	577.68	Joback Method
cpg	383.56	J/molxK	607.67	Joback Method
cpg	394.94	J/molxK	637.65	Joback Method
cpg	405.72	J/molxK	667.63	Joback Method
cpg	415.94	J/molxK	697.62	Joback Method
cpg	425.62	J/molxK	727.60	Joback Method

dvisc	0.0282170	Paxs	289.36	Joback Method
dvisc	0.0061539	Paxs	332.42	Joback Method
dvisc	0.0019031	Paxs	375.47	Joback Method
dvisc	0.0007493	Paxs	418.53	Joback Method
dvisc	0.0003510	Paxs	461.59	Joback Method
dvisc	0.0001872	Paxs	504.64	Joback Method
dvisc	0.0001102	Paxs	547.70	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.56981e+01
Coeff. B	-4.68506e+03
Coeff. C	-8.13040e+01
Temperature range (K), min.	385.32
Temperature range (K), max.	532.37

## Sources

<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3302101&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3302101&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log10 of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>p<sub>c</sub>:</b>	Critical Pressure
<b>p<sub>vap</sub>:</b>	Vapor pressure
<b>r<sub>inpol</sub>:</b>	Non-polar retention indices
<b>t<sub>b</sub>:</b>	Normal Boiling Point Temperature
<b>t<sub>c</sub>:</b>	Critical Temperature
<b>t<sub>f</sub>:</b>	Normal melting (fusion) point
<b>v<sub>c</sub>:</b>	Critical Volume

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