

Tetradecanoic acid

Other names:	1-Tridecanecarboxylic acid Acide Myristique Crodacid Emery 655 Hydrofol acid 1495 Hystrene 9014 Hystrene 9514 Myristic acid Myristic acid (tetradecanoic acid) N-MYRISTIC ACID NSC 5028 Neo-Fat 14 Philacid 1400 Prifac 2942 Prifrac 2942 Tetradecanoic (Myristic) acid Tetradecanoic acid (=Myristic acid) Tetradecanoic acid (myristic acid) Univol U 316S n-Tetradecan-1-oic acid n-Tetradecanoic acid n-Tetradecoic acid
Inchi:	InChI=1S/C14H28O2/c1-2-3-4-5-6-7-8-9-10-11-12-13-14(15)16/h2-13H2,1H3,(H,15,16)
InchiKey:	TUNFSRHWOTWDNC-UHFFFAOYSA-N
Formula:	C14H28O2
SMILES:	CCCCCCCCCCCC(=O)O
Mol. weight [g/mol]:	228.37
CAS:	544-63-8

Physical Properties

Property code	Value	Unit	Source
chl	-8676.70 ± 1.40	kJ/mol	NIST Webbook
chs	-8675.90 ± 9.60	kJ/mol	NIST Webbook
gf	-198.74	kJ/mol	Joback Method
hf	-597.10	kJ/mol	Joback Method

hfus	44.75	kJ/mol	Solid-liquid phase equilibria of (n-octadecane with myristic, and palmitic acid) binary mixtures used as phase change materials (PCMs)
hfus	43.70	kJ/mol	Solid-Liquid Equilibrium of Binary Systems Containing Fatty Acids and Fatty Alcohols Using Differential Scanning Calorimetry
hsub	169.00 ± 9.00	kJ/mol	NIST Webbook
hvap	70.18	kJ/mol	Joback Method
log10ws	-5.33		Aqueous Solubility Prediction Method
logp	4.772		Crippen Method
mcvol	215.560	ml/mol	McGowan Method
pc	1644.20 ± 85.00	kPa	NIST Webbook
pt	2.13e-05 ± 1.07e-05	kPa	NIST Webbook
rinpol	1769.00		NIST Webbook
rinpol	1762.00		NIST Webbook
rinpol	1786.00		NIST Webbook
rinpol	1745.00		NIST Webbook
rinpol	1750.00		NIST Webbook
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ripol	2713.00		NIST Webbook
ripol	2701.00		NIST Webbook
tb	665.77	K	Joback Method
tc	765.19 ± 3.00	K	NIST Webbook
tf	328.88	K	The solid liquid phase diagrams of binary mixtures of even saturated fatty acids differing by six carbon atoms
tf	327.82	K	Aqueous Solubility Prediction Method
tf	327.50	K	High pressure solid liquid equilibria of fatty acids
tf	327.03	K	Measurement and PC-SAFT modeling of solid-liquid equilibrium of deep eutectic solvents of quaternary ammonium chlorides and carboxylic acids
tf	329.05	K	Solid-liquid phase equilibrium diagrams of binary mixtures containing fatty acids, fatty alcohol compounds and tripalmitin using differential scanning calorimetry

tf	327.50	K	Study of the Effect of Pressure on Melting Behavior of Saturated Fatty Acids in Liquid or Supercritical Carbon Dioxide
tf	326.15	K	Preparation of paraffin and fatty acid phase changing nanoemulsions for heat transfer
tf	327.85	K	Prediction of the properties of eutectic fatty acid phase change materials
tf	327.60	K	Solid Liquid Equilibria in the Binary Systems of Saturated Fatty Acids or Triglycerides (C12 to C18) + Hexadecane
tf	327.48	K	Solid-Liquid Equilibrium of Binary Fatty Acid Mixtures
tt	327.32 ± 0.02	K	NIST Webbook
tt	327.05 ± 0.50	K	NIST Webbook
vc	0.845	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	676.89	J/mol×K	804.44	Joback Method
cpg	608.69	J/mol×K	665.77	Joback Method
cpg	623.60	J/mol×K	693.50	Joback Method
cpg	637.85	J/mol×K	721.24	Joback Method
cpg	651.47	J/mol×K	748.97	Joback Method
cpg	664.48	J/mol×K	776.71	Joback Method
cpg	688.73	J/mol×K	832.18	Joback Method
cps	523.00	J/mol×K	298.00	NIST Webbook
cps	432.01	J/mol×K	298.15	NIST Webbook
dvisc	0.0002574	Paxs	512.03	Joback Method
dvisc	0.0005619	Paxs	460.78	Joback Method
dvisc	0.0000798	Paxs	614.52	Joback Method
dvisc	0.0000509	Paxs	665.77	Joback Method
dvisc	0.0014913	Paxs	409.54	Joback Method
dvisc	0.0052331	Paxs	358.29	Joback Method
dvisc	0.0001359	Paxs	563.28	Joback Method
hfust	45.10	kJ/mol	327.00	NIST Webbook
hfust	45.10	kJ/mol	327.00	NIST Webbook
hfust	45.00	kJ/mol	326.50	NIST Webbook

hfust	40.10	kJ/mol	326.60	NIST Webbook
hfust	45.75	kJ/mol	326.20	NIST Webbook
hfust	36.28	kJ/mol	317.00	NIST Webbook
hsubt	139.70 ± 3.80	kJ/mol	318.50	NIST Webbook
hsubt	140.00 ± 4.00	kJ/mol	311.85	NIST Webbook
hvapt	100.40	kJ/mol	421.00	NIST Webbook
hvapt	91.60	kJ/mol	511.00	NIST Webbook
hvapt	104.10 ± 2.00	kJ/mol	348.50	NIST Webbook
hvapt	88.90	kJ/mol	455.00	NIST Webbook
hvapt	111.20	kJ/mol	298.00	Vapor Pressures and Vaporization, Sublimation, and Fusion Enthalpies of Some Fatty Acids
pvap	1.33	kPa	464.60	Vapor pressure data for fatty acids obtained using an adaptation of the DSC technique
pvap	5.33	kPa	497.20	Vapor pressure data for fatty acids obtained using an adaptation of the DSC technique
pvap	6.67	kPa	503.10	Vapor pressure data for fatty acids obtained using an adaptation of the DSC technique
pvap	8.00	kPa	508.20	Vapor pressure data for fatty acids obtained using an adaptation of the DSC technique
pvap	9.33	kPa	512.60	Vapor pressure data for fatty acids obtained using an adaptation of the DSC technique
pvap	2.67	kPa	480.80	Vapor pressure data for fatty acids obtained using an adaptation of the DSC technique
pvap	4.00	kPa	489.50	Vapor pressure data for fatty acids obtained using an adaptation of the DSC technique

rho1	806.70	kg/m3	408.20	Density, Viscosity, and Thermal Conductivity of Eight Carboxylic Acids from (290.3 to 473.4) K
rho1	846.90	kg/m3	352.10	Density, Viscosity, and Thermal Conductivity of Eight Carboxylic Acids from (290.3 to 473.4) K
rho1	835.60	kg/m3	368.20	Density, Viscosity, and Thermal Conductivity of Eight Carboxylic Acids from (290.3 to 473.4) K
rho1	791.70	kg/m3	427.90	Density, Viscosity, and Thermal Conductivity of Eight Carboxylic Acids from (290.3 to 473.4) K
rho1	758.00	kg/m3	473.20	Density, Viscosity, and Thermal Conductivity of Eight Carboxylic Acids from (290.3 to 473.4) K
rho1	769.30	kg/m3	458.20	Density, Viscosity, and Thermal Conductivity of Eight Carboxylic Acids from (290.3 to 473.4) K
rho1	780.90	kg/m3	443.20	Density, Viscosity, and Thermal Conductivity of Eight Carboxylic Acids from (290.3 to 473.4) K
rho1	821.40	kg/m3	388.20	Density, Viscosity, and Thermal Conductivity of Eight Carboxylic Acids from (290.3 to 473.4) K
rho1	861.30	kg/m3	332.10	Density, Viscosity, and Thermal Conductivity of Eight Carboxylic Acids from (290.3 to 473.4) K

Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	523.20	K	13.30	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.57451e+01
Coeff. B	-5.46750e+03
Coeff. C	-1.07769e+02
Temperature range (K), min.	461.48
Temperature range (K), max.	631.79

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	2.20853e+02
Coeff. B	-2.15189e+04
Coeff. C	-2.86975e+01
Coeff. D	9.02023e-06
Temperature range (K), min.	327.55
Temperature range (K), max.	756.00

Sources

KDB Vapor Pressure Data: <https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=948>

Aqueous Solubility Prediction Method: <http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx>

Crippen Method: <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Measurement and PC-SAFT modeling of solid-liquid equilibrium of deep eutectic solvents of quaternary ammonium chlorides and carboxylic acids: <https://www.doi.org/10.1016/j.fluid.2017.04.007>

Prediction of the properties of eutectic fatty acid phase change materials: Solid-liquid phase equilibria of (n-octadecane with myristic, and lauric acid) binary mixtures used as phase change materials (PCMs): Activity coefficient at infinite dilution measurements for organic solutes Solubilities of Dodecanols and Tetradecanoic Acids in Supercritical CO₂ without and with the presence of fatty acids: Solubilities of Fatty Acids and Triglycerides in 1-Bromopropane: Preparation of paraffin and fatty acid phase changing nanoemulsions for heat transfer: Vapor pressure data for fatty acids obtained using an adaptation of the Solid-Liquid phase equilibrium diagrams of binary mixtures containing fatty acids: Properties of some compounds of interest to the edible oil industry: Solubilities and Diffusivities of Binary Systems of n-Heptane (Fatty Acids of Molecular Weight (C₁₈) + Hexadecane: Density, Viscosity, and Thermal Conductivity of Eight Carboxylic Acids NIST Webbook (73.4) K: <https://www.doi.org/10.1016/j.tca.2017.12.024>
<https://www.doi.org/10.1016/j.jct.2016.05.008>
<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>
<https://www.doi.org/10.1016/j.jct.2012.06.009>
<https://www.doi.org/10.1021/je800544a>
<https://www.doi.org/10.1016/j.fluid.2007.02.007>
<https://www.doi.org/10.1021/je201181k>
<https://www.doi.org/10.1016/j.tca.2014.12.020>
<https://www.doi.org/10.1016/j.tca.2012.07.034>
<https://www.doi.org/10.1016/j.fluid.2019.05.020>
<https://www.doi.org/10.1016/j.jct.2017.06.012>
<https://www.doi.org/10.1021/acs.jced.6b00355>
https://en.wikipedia.org/wiki/Joback_method
<https://www.doi.org/10.1021/acs.jced.5b00971>
<http://webbook.nist.gov/cgi/cbook.cgi?ID=C544638&Units=SI>
<https://www.doi.org/10.1021/je400260c>
<https://www.doi.org/10.1021/acs.jced.8b01006>
<https://www.doi.org/10.1021/je060146z>
<https://www.doi.org/10.1016/j.jct.2009.07.008>
<https://www.doi.org/10.1021/je300902c>
<https://www.doi.org/10.1016/j.tca.2009.06.018>
<http://link.springer.com/article/10.1007/BF02311772>

Study of the Effect of Pressure on Melting Behavior of Saturated Fatty Acids: Solid-Liquid Equilibria of Binary Systems Containing Fatty Acids and Ethyl Alcohol: Estimating the Primary Fatty Acid Molar Fractions: Experimental determination of the (vapor + liquid) equilibrium data of Vapor Pressures and Vaporization, Sublimation, and Fusion Enthalpies of Some Fatty Acids: Binary phase diagrams of binary mixtures of even saturated fatty acids containing six carbon atoms: <https://www.doi.org/10.1021/je060146z>
<https://www.doi.org/10.1016/j.jct.2009.07.008>
<https://www.doi.org/10.1021/je300902c>
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Legend

chl:	Standard liquid enthalpy of combustion
chs:	Standard solid enthalpy of combustion
cpg:	Ideal gas heat capacity
cps:	Solid phase heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hfust:	Enthalpy of fusion at a given temperature
hsub:	Enthalpy of sublimation at standard conditions
hsubt:	Enthalpy of sublimation at a given temperature
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume

pc:	Critical Pressure
pt:	Triple Point Pressure
pvap:	Vapor pressure
rho:	Liquid Density
rinpol:	Non-polar retention indices
ripol:	Polar retention indices
sfust:	Entropy of fusion at a given temperature
tb:	Normal Boiling Point Temperature
tbrp:	Boiling point at reduced pressure
tc:	Critical Temperature
tf:	Normal melting (fusion) point
tt:	Triple Point Temperature
vc:	Critical Volume

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