

# 2-Methoxy-5-nitrobenzotrifluoride

<b>Other names:</b>	5-Nitro-2-methoxybenzotrifluoride Benzene, 1-methoxy-4-nitro-2-(trifluoromethyl)- Anisole, 4-nitro-2-(trifluoromethyl)- 4-nitro-2-(trifluoromethyl)anisole
<b>Inchi:</b>	InChI=1S/C8H6F3NO3/c1-15-7-3-2-5(12(13)14)4-6(7)8(9,10)11/h2-4H,1H3
<b>InchiKey:</b>	KGFADEJSZXEVMC-UHFFFAOYSA-N
<b>Formula:</b>	C8H6F3NO3
<b>SMILES:</b>	COc1ccc([N+](=O)[O-])cc1C(F)(F)F
<b>Mol. weight [g/mol]:</b>	221.13
<b>CAS:</b>	654-76-2

## Physical Properties

Property code	Value	Unit	Source
gf	-541.41	kJ/mol	Joback Method
hf	-734.92	kJ/mol	Joback Method
hfus	24.11	kJ/mol	Joback Method
hvap	52.26	kJ/mol	Joback Method
log10ws	-3.35		Crippen Method
logp	2.622		Crippen Method
mcvol	128.420	ml/mol	McGowan Method
pc	3065.95	kPa	Joback Method
tb	587.92	K	Joback Method
tc	809.18	K	Joback Method
tf	401.41	K	Joback Method
vc	0.518	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	315.13	J/molxK	587.92	Joback Method
cpg	325.67	J/molxK	624.80	Joback Method
cpg	335.44	J/molxK	661.67	Joback Method
cpg	344.46	J/molxK	698.55	Joback Method
cpg	352.78	J/molxK	735.42	Joback Method

cpg	360.43	J/mol×K	772.30	Joback Method
cpg	367.44	J/mol×K	809.18	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C654762&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C654762&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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