

# Fenothiocarb

<b>Other names:</b>	Carbamothioic acid, N,N-dimethyl-, S-(4-phenoxybutyl) ester
<b>Inchi:</b>	InChI=1S/C13H19NO2S/c1-14(2)13(15)17-11-7-6-10-16-12-8-4-3-5-9-12/h3-5,8-9H,6-7,1
<b>InchiKey:</b>	HMIBKHHNXANVHR-UHFFFAOYSA-N
<b>Formula:</b>	C13H19NO2S
<b>SMILES:</b>	CN(C)C(=O)SCCCCOc1ccccc1
<b>Mol. weight [g/mol]:</b>	253.36
<b>CAS:</b>	62850-32-2

## Physical Properties

Property code	Value	Unit	Source
gf	80.97	kJ/mol	Joback Method
hf	-210.52	kJ/mol	Joback Method
hfus	33.40	kJ/mol	Joback Method
hvap	64.82	kJ/mol	Joback Method
log10ws	-3.93		Estimated Solubility Method
log10ws	-3.93		Aqueous Solubility Prediction Method
logp	3.260		Crippen Method
mcvol	204.040	ml/mol	McGowan Method
pc	2324.78	kPa	Joback Method
rinpol	2136.00		NIST Webbook
rinpol	2136.00		NIST Webbook
tb	681.03	K	Joback Method
tc	896.92	K	Joback Method
tf	313.65	K	Aqueous Solubility Prediction Method
vc	0.751	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	538.79	J/molxK	681.03	Joback Method
cpg	554.50	J/molxK	717.01	Joback Method
cpg	569.15	J/molxK	752.99	Joback Method

cpg	582.77	J/mol×K	788.97	Joback Method
cpg	595.40	J/mol×K	824.95	Joback Method
cpg	607.05	J/mol×K	860.93	Joback Method
cpg	617.78	J/mol×K	896.92	Joback Method

## Sources

**Aqueous Solubility Prediction Method:** <http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa>

**Estimated Solubility Method:** [http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl\\_file/ci034243xsi20040112\\_053635.txt](http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt)

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C62850322&Units=SI>

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci990307I>

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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