

# Ethanol, 2,2'-oxybis-, dinitrate

<b>Other names:</b>	Diethylene glycol, dinitrate Diglycol dinitrate Dinitrodiglycol Oxydiethylene nitrate 2,2'-Oxydiethanol dinitrate Digol dinitrate Bis(hydroxyaethyl)-aether-dinitrat Diethylenglykoldinitrate Diglycoldinitraat Diglykoldinitrat Di(hydroxyethyl) ether dinitrate Dinitrate de diethylene-glycol Dinitrodiglicol Dinitrodiglykol DEGDN Ethanol, 2,2'-oxybis-, 1,1'-dinitrate oxydiethylene dinitrate
<b>Inchi:</b>	InChI=1S/C4H8N2O7/c7-5(8)12-3-1-11-2-4-13-6(9)10/h1-4H2
<b>InchiKey:</b>	LYAGTVMJGHTIDH-UHFFFAOYSA-N
<b>Formula:</b>	C4H8N2O7
<b>SMILES:</b>	O=[N+](O-)[O-]OCCOCCO[N+](=O)[O-]
<b>Mol. weight [g/mol]:</b>	196.12
<b>CAS:</b>	693-21-0

## Physical Properties

Property code	Value	Unit	Source
chl	-2261.00	kJ/mol	NIST Webbook
chl	-2291.00	kJ/mol	NIST Webbook
chl	-2297.60	kJ/mol	NIST Webbook
chl	-2291.00	kJ/mol	NIST Webbook
gf	-261.10	kJ/mol	Joback Method
hf	-544.07	kJ/mol	Joback Method
hfl	-451.00	kJ/mol	NIST Webbook
hfs	-476.10	kJ/mol	NIST Webbook
hfus	32.40	kJ/mol	Joback Method
hvap	64.91	kJ/mol	Joback Method
log10ws	-0.91		Crippen Method

logp	-0.580		Crippen Method
mvol	119.670	ml/mol	McGowan Method
pc	3768.41	kPa	Joback Method
rinpol	1351.13		NIST Webbook
rinpol	1357.28		NIST Webbook
rinpol	1332.85		NIST Webbook
rinpol	1344.19		NIST Webbook
tb	661.86	K	Joback Method
tc	889.13	K	Joback Method
tf	488.75	K	Joback Method
vc	0.477	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	347.11	J/mol×K	813.37	Joback Method
cpg	353.93	J/mol×K	851.25	Joback Method
cpg	313.45	J/mol×K	661.86	Joback Method
cpg	322.77	J/mol×K	699.74	Joback Method
cpg	331.51	J/mol×K	737.62	Joback Method
cpg	339.63	J/mol×K	775.49	Joback Method
cpg	360.04	J/mol×K	889.13	Joback Method
hfust	25.40	kJ/mol	276.50	NIST Webbook
hvapt	94.30	kJ/mol	313.00	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C693210&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C693210&amp;Units=SI</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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