

# Tricyclo[3.3.1.1(3,7)]decane, 2-nitro-

<b>Other names:</b>	2-Nitroadamantane
<b>Inchi:</b>	InChI=1S/C10H15NO2/c12-11(13)10-8-2-6-1-7(4-8)5-9(10)3-6/h6-10H,1-5H2
<b>InchiKey:</b>	ZFSCSQSQXFIKKH-UHFFFAOYSA-N
<b>Formula:</b>	C10H15NO2
<b>SMILES:</b>	O=[N+]([O-])C1C2CC3CC(C2)CC1C3
<b>Mol. weight [g/mol]:</b>	181.23
<b>CAS:</b>	54564-31-7

## Physical Properties

Property code	Value	Unit	Source
chs	-5841.00 ± 2.20	kJ/mol	NIST Webbook
gf	223.60	kJ/mol	Joback Method
hf	-179.80 ± 3.20	kJ/mol	NIST Webbook
hfs	-237.80 ± 2.20	kJ/mol	NIST Webbook
hfus	27.46	kJ/mol	Joback Method
hsub	58.00 ± 2.30	kJ/mol	NIST Webbook
hvap	53.74	kJ/mol	Joback Method
log10ws	-2.92		Crippen Method
logp	2.088		Crippen Method
mcvol	136.600	ml/mol	McGowan Method
pc	3028.94	kPa	Joback Method
tb	595.19	K	Joback Method
tc	839.06	K	Joback Method
tf	387.89	K	Joback Method
vc	0.538	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	395.47	J/molxK	595.19	Joback Method
cpg	414.67	J/molxK	635.83	Joback Method
cpg	432.36	J/molxK	676.48	Joback Method
cpg	448.69	J/molxK	717.12	Joback Method
cpg	463.79	J/molxK	757.77	Joback Method

cpg	477.79	J/mol×K	798.41	Joback Method
cpg	490.84	J/mol×K	839.06	Joback Method
hfust	4.23	kJ/mol	452.20	NIST Webbook
hsubt	58.00 ± 2.30	kJ/mol	349.50	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C54564317&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C54564317&amp;Units=SI</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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