

# Tetradecyl formate

<b>Inchi:</b>	InChI=1S/C15H30O2/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-17-15-16/h15H,2-14H2,1H3
<b>InchiKey:</b>	SJBKZSQQFOWRQX-UHFFFAOYSA-N
<b>Formula:</b>	C15H30O2
<b>SMILES:</b>	CCCCCCCCCCCCCOC=O
<b>Mol. weight [g/mol]:</b>	242.40
<b>CAS:</b>	5451-63-8

## Physical Properties

Property code	Value	Unit	Source
gf	-129.10	kJ/mol	Joback Method
hf	-570.73	kJ/mol	Joback Method
hfus	38.08	kJ/mol	Joback Method
hvap	58.11	kJ/mol	Joback Method
log10ws	-4.96		Crippen Method
logp	4.861		Crippen Method
mvol	229.650	ml/mol	McGowan Method
pc	1478.15	kPa	Joback Method
rinpol	1710.00		NIST Webbook
rinpol	1710.00		NIST Webbook
tb	613.68	K	Joback Method
tc	778.21	K	Joback Method
tf	323.04	K	Joback Method
vc	0.910	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	620.17	J/molxK	613.68	Joback Method
cpg	698.98	J/molxK	750.79	Joback Method
cpg	684.56	J/molxK	723.36	Joback Method
cpg	669.47	J/molxK	695.94	Joback Method
cpg	653.72	J/molxK	668.52	Joback Method
cpg	637.29	J/molxK	641.10	Joback Method
cpg	712.77	J/molxK	778.21	Joback Method

dvisc	0.0001513	Paxs	613.68	Joback Method
dvisc	0.0002011	Paxs	565.24	Joback Method
dvisc	0.0002820	Paxs	516.80	Joback Method
dvisc	0.0004239	Paxs	468.36	Joback Method
dvisc	0.0007002	Paxs	419.92	Joback Method
dvisc	0.0013183	Paxs	371.48	Joback Method
dvisc	0.0030006	Paxs	323.04	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	2.07562e+01
Coeff. B	-7.25030e+03
Coeff. C	-1.09104e+02
Temperature range (K), min.	463.32
Temperature range (K), max.	578.54

## Sources

The Yaws Handbook of Vapor Pressure:  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Joback Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

Joback Method:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=R543825&Units=SI>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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