

# Barbituric acid, 5-allyl-5-neopentyl-

**Other names:**

Nealbarbital  
5-Allyl-5-neopentylbarbituric acid  
2,4,6(1H,3H,5H)-Pyrimidinetrione, 5-(2,2-dimethylpropyl)-5-(2-propenyl)-  
Allylneopentylbarbituric acid  
Censedal  
Nealbarbitone  
Nepental  
Nevental  
5-Neopentyl-5-allylbarbituric acid  
5055  
5-(2,2-Dimethylpropyl)-5-(2-propenyl)-2,4,6(1H,3H,5H)-pyrimidinetrione  
Neallymal  
5-Allyl-5-(2,2-dimethylpropyl)barbituric acid

**Inchi:**

InChI=1S/C12H18N2O3/c1-5-6-12(7-11(2,3)4)8(15)13-10(17)14-9(12)16/h5H,1,6-7H2,2-

**InchiKey:**

YHKPTICJRUESOY-UHFFFAOYSA-N

**Formula:**

C12H18N2O3

**SMILES:**

C=CCC1(CC(C)(C)C)C(=O)NC(=O)NC1=O

**Mol. weight [g/mol]:**

238.28

**CAS:**

561-83-1

## Physical Properties

Property code	Value	Unit	Source
gf	-32.55	kJ/mol	Joback Method
hf	-442.25	kJ/mol	Joback Method
hfus	21.39	kJ/mol	Joback Method
hvap	65.88	kJ/mol	Joback Method
log10ws	-2.79		Crippen Method
logp	1.351		Crippen Method
mcvol	189.450	ml/mol	McGowan Method
pc	2758.46	kPa	Joback Method
rinpol	1720.00		NIST Webbook
rinpol	1720.00		NIST Webbook
tb	787.76	K	Joback Method
tc	1048.64	K	Joback Method
tf	671.66	K	Joback Method
vc	0.704	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	581.76	J/mol×K	787.76	Joback Method
cpg	600.44	J/mol×K	831.24	Joback Method
cpg	618.07	J/mol×K	874.72	Joback Method
cpg	634.73	J/mol×K	918.20	Joback Method
cpg	650.46	J/mol×K	961.68	Joback Method
cpg	665.33	J/mol×K	1005.16	Joback Method
cpg	679.41	J/mol×K	1048.64	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C561831&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C561831&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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