

# 1-Butanamine, 3-methyl-N-ethylidene

<b>Other names:</b>	Butanamine, 3-methyl-N-ethylidene
<b>Inchi:</b>	InChI=1S/C7H15N/c1-4-8-6-5-7(2)3/h4,7H,5-6H2,1-3H3
<b>InchiKey:</b>	SHFUPXUCTHPZGG-UHFFFAOYSA-N
<b>Formula:</b>	C7H15N
<b>SMILES:</b>	CC=NCCC(C)C
<b>Mol. weight [g/mol]:</b>	113.20

## Physical Properties

Property code	Value	Unit	Source
hf	-110.87	kJ/mol	Joback Method
hvap	34.10	kJ/mol	Joback Method
log10ws	-1.68		Crippen Method
logp	2.123		Crippen Method
mcvol	115.170	ml/mol	McGowan Method
pc	2532.82	kPa	Joback Method
rinpol	792.00		NIST Webbook
rinpol	790.00		NIST Webbook
rinpol	790.00		NIST Webbook
rinpol	797.00		NIST Webbook
rinpol	792.00		NIST Webbook
ripol	974.00		NIST Webbook
ripol	974.00		NIST Webbook
tb	435.80	K	Joback Method
tc	626.17	K	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R45792&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R45792&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

# Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>ripolar:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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