

# 2-Ethylbutyric acid, 2-naphthyl ester

<b>Inchi:</b>	InChI=1S/C16H18O2/c1-3-12(4-2)16(17)18-15-10-9-13-7-5-6-8-14(13)11-15/h5-12H,3-4H
<b>InchiKey:</b>	JDKZXLBYHYCZKA-UHFFFAOYSA-N
<b>Formula:</b>	C16H18O2
<b>SMILES:</b>	CCC(CC)C(=O)Oc1ccc2ccccc2c1
<b>Mol. weight [g/mol]:</b>	242.31

## Physical Properties

Property code	Value	Unit	Source
gf	56.91	kJ/mol	Joback Method
hf	-207.52	kJ/mol	Joback Method
hfus	27.13	kJ/mol	Joback Method
hvap	64.56	kJ/mol	Joback Method
log10ws	-5.02		Crippen Method
logp	4.181		Crippen Method
mcvol	200.520	ml/mol	McGowan Method
pc	2185.64	kPa	Joback Method
rinpola	1941.00		NIST Webbook
tb	691.97	K	Joback Method
tc	914.36	K	Joback Method
tf	398.88	K	Joback Method
vc	0.763	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	537.59	J/molxK	691.97	Joback Method
cpg	606.69	J/molxK	877.29	Joback Method
cpg	594.78	J/molxK	840.23	Joback Method
cpg	581.97	J/molxK	803.16	Joback Method
cpg	568.21	J/molxK	766.10	Joback Method
cpg	553.44	J/molxK	729.03	Joback Method
cpg	617.76	J/molxK	914.36	Joback Method
dvisc	0.0001999	Paxs	691.97	Joback Method
dvisc	0.0002464	Paxs	643.12	Joback Method

dvisc	0.0003145	Paxs	594.27	Joback Method
dvisc	0.0004192	Paxs	545.42	Joback Method
dvisc	0.0005914	Paxs	496.58	Joback Method
dvisc	0.0008993	Paxs	447.73	Joback Method
dvisc	0.0015154	Paxs	398.88	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369890&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369890&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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