

# O-methyloxime pentadecanal

Inchi:	InChI=1S/C16H33NO/c1-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-2/h16H,3-15H2,1-2H
InchiKey:	AUZWSQVZJNSAGW-UHFFFAOYSA-N
Formula:	C16H33NO
SMILES:	CCCCCCCCCC=NOCC
Mol. weight [g/mol]:	255.44

## Physical Properties

Property code	Value	Unit	Source
hf	-423.57	kJ/mol	Joback Method
hvap	56.93	kJ/mol	Joback Method
log10ws	-5.76		Crippen Method
logp	5.710		Crippen Method
mcvol	247.850	ml/mol	McGowan Method
pc	1208.99	kPa	Joback Method
rinpol	1776.00		NIST Webbook
tb	664.58	K	Joback Method
tc	837.77	K	Joback Method

## Sources

Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R213717&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R213717&amp;Units=SI</a>

## Legend

hf:	Enthalpy of formation at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l

<b>logP:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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