

# 2-Furanmethanol, tetrahydro-, acetate

<b>Other names:</b>	Furfuryl alcohol, tetrahydro-, acetate Tetrahydrofurfuryl acetate 2-Acetoxymethyloxolane Tetrahydro-2-furanmethanol acetate Tetrahydro-2-furanmethyl acetate Acetic acid, tetrahydrofurfuryl ester
<b>Inchi:</b>	InChI=1S/C7H12O3/c1-6(8)10-5-7-3-2-4-9-7/h7H,2-5H2,1H3
<b>InchiKey:</b>	AAQDYFFAFXGBFZ-UHFFFAOYSA-N
<b>Formula:</b>	C7H12O3
<b>SMILES:</b>	CC(=O)OCC1CCCO1
<b>Mol. weight [g/mol]:</b>	144.17
<b>CAS:</b>	637-64-9

## Physical Properties

Property code	Value	Unit	Source
gf	-275.43	kJ/mol	Joback Method
hf	-504.13	kJ/mol	Joback Method
hfus	18.59	kJ/mol	Joback Method
hvap	45.10	kJ/mol	Joback Method
log10ws	-0.71		Crippen Method
logp	0.728		Crippen Method
mcvol	111.940	ml/mol	McGowan Method
pc	3572.80	kPa	Joback Method
rinpol	1055.00		NIST Webbook
rinpol	1098.00		NIST Webbook
rinpol	1055.00		NIST Webbook
rinpol	1098.00		NIST Webbook
tb	478.08	K	Joback Method
tc	683.59	K	Joback Method
tf	278.28	K	Joback Method
vc	0.413	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	250.27	J/mol×K	478.08	Joback Method
cpg	263.89	J/mol×K	512.33	Joback Method
cpg	276.85	J/mol×K	546.58	Joback Method
cpg	289.16	J/mol×K	580.84	Joback Method
cpg	300.82	J/mol×K	615.09	Joback Method
cpg	311.86	J/mol×K	649.34	Joback Method
cpg	322.28	J/mol×K	683.59	Joback Method
dvisc	0.0034190	Paxs	278.28	Joback Method
dvisc	0.0019399	Paxs	311.58	Joback Method
dvisc	0.0012280	Paxs	344.88	Joback Method
dvisc	0.0008425	Paxs	378.18	Joback Method
dvisc	0.0006144	Paxs	411.48	Joback Method
dvisc	0.0004697	Paxs	444.78	Joback Method
dvisc	0.0003728	Paxs	478.08	Joback Method

## Sources

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C637649&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307i>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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