

# 4a«alpha»,4b«beta»-Gibbane-1«alpha»,10«beta»-

Other names: Dimethyl 4a-formyl-1-methyl-8-methylenegibbane-1,10-dicarboxylate, (4a«alpha»,4b«beta»,10«beta»,10«beta»)Gibbane-1,10-dicarboxylate, GA24, Me

**acid, 4a-formyl-1-methyl-8-methylene-, dimethyl ester**

**InchiKey:** CDZKHHKZGGMPOC-UHFFFAOYSA-N

**Formula:** C22H30O5

**SMILES:** C=C1CC23CC1CCC2C1(C=O)CCCC(C)(C(=O)OC)C1C3C(=O)OC

**Mol. weight [g/mol]:** 374.47

**CAS:** 19427-33-9

## Physical Properties

Property code	Value	Unit	Source
gf	-212.82	kJ/mol	Joback Method
hf	-730.93	kJ/mol	Joback Method
hfus	30.00	kJ/mol	Joback Method
hvap	85.55	kJ/mol	Joback Method
log10ws	-3.78		Crippen Method
logp	3.317		Crippen Method
mcvol	289.550	ml/mol	McGowan Method
pc	1582.23	kPa	Joback Method
rinpol	2443.00		NIST Webbook
rinpol	2469.00		NIST Webbook
rinpol	2440.00		NIST Webbook
rinpol	2441.00		NIST Webbook
rinpol	2441.00		NIST Webbook
rinpol	2477.00		NIST Webbook
rinpol	2444.00		NIST Webbook
rinpol	2443.00		NIST Webbook
rinpol	2444.00		NIST Webbook
rinpol	2453.00		NIST Webbook
rinpol	2444.00		NIST Webbook
rinpol	2453.00		NIST Webbook
rinpol	2453.00		NIST Webbook
rinpol	2453.00		NIST Webbook
rinpol	2451.00		NIST Webbook
rinpol	2448.00		NIST Webbook
rinpol	2454.00		NIST Webbook
tb	929.64	K	Joback Method

tc	1164.39	K	Joback Method
tf	657.88	K	Joback Method
vc	1.111	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1041.61	J/mol×K	929.64	Joback Method
cpg	1070.38	J/mol×K	968.77	Joback Method
cpg	1100.77	J/mol×K	1007.89	Joback Method
cpg	1133.19	J/mol×K	1047.02	Joback Method
cpg	1168.10	J/mol×K	1086.14	Joback Method
cpg	1205.91	J/mol×K	1125.27	Joback Method
cpg	1247.07	J/mol×K	1164.39	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C19427339&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C19427339&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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