

# Benzenamine, 2,4,6-tris(1,1-dimethylethyl)-

**Other names:** 2,4,6-tri-t-Butylaniline; 2,4,6-tri-tert-Butylaniline; Aniline, 2,4,6-tri-tert-butyl-; Aniline, tri-tert-butyl-

**InChI:** InChI=1S/C18H31N/c1-16(2,3)12-10-13(17(4,5)6)15(19)14(11-12)18(7,8)9/h10-11H,19H2,1-9H3

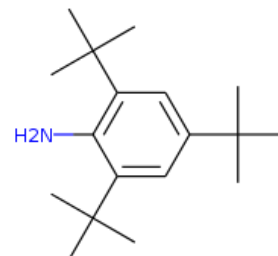
**InChI Key:** REJGDSCBQPJPQT-UHFFFAOYSA-N

**Formula:** C18H31N

**SMILES:** CC(C)(C)c1cc(C(C)(C)C)c(N)c(C(C)(C)C)c1

**Molecular Weight:** 261.45

**CAS:** 961-38-6



## Physical Properties

Property	Value	Unit	Source
$\Delta_f G^\circ$	259.17	kJ/mol	Joback Method
$\Delta_f H^\circ_{\text{gas}}$	-205.19	kJ/mol	Joback Method
$\Delta_{\text{fus}} H^\circ$	18.21	kJ/mol	Joback Method
$\Delta_{\text{sub}} H^\circ$	92.50 ± 1.10	kJ/mol	NIST Webbook
$\Delta_{\text{vap}} H^\circ$	66.68	kJ/mol	Joback Method
IE	6.90	eV	NIST Webbook
IE	7.30	eV	NIST Webbook
$\log P_{\text{oct/wat}}$	5.16		Crippen Method
$P_c$	1542.71	kPa	Joback Method
$T_{\text{boil}}$	715.70	K	Joback Method
$T_c$	941.27	K	Joback Method
$T_{\text{fus}}$	447.12	K	Joback Method
$V_c$	0.93	m <sup>3</sup> /kg-mol	Joback Method

## Temperature Dependent Properties

Property	Value	Unit	Temperature (K)	Source
$C_{p,gas}$	736.39	J/mol×K	715.7	Joback Method
$\Delta_{fus}H$	19.38	kJ/mol	426.4	NIST Webbook

## Sources

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**NIST Webbook:** [http://webbook.nist.gov/cgi/inchi/InChI=1S/C18H31N/c1-16\(2,3\)12-10-13\(17\(4,5\)6\)15\(19\)14\(11-12\)18\(7,8\)9/h10-11H,19H2,1-9H3](http://webbook.nist.gov/cgi/inchi/InChI=1S/C18H31N/c1-16(2,3)12-10-13(17(4,5)6)15(19)14(11-12)18(7,8)9/h10-11H,19H2,1-9H3)

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

## Legend

$C_{p,gas}$ : Ideal gas heat capacity (J/mol×K).

$\Delta_f G^\circ$ : Standard Gibbs free energy of formation (kJ/mol).

$\Delta_f H^\circ_{gas}$ : Enthalpy of formation at standard conditions (kJ/mol).

$\Delta_{fus} H^\circ$ : Enthalpy of fusion at standard conditions (kJ/mol).

$\Delta_{fus} H$ : Enthalpy of fusion at a given temperature (kJ/mol).

$\Delta_{sub} H^\circ$ : Enthalpy of sublimation at standard conditions (kJ/mol).

$\Delta_{vap} H^\circ$ : Enthalpy of vaporization at standard conditions (kJ/mol).

**IE:** Ionization energy (eV).

**logP<sub>oct/wat</sub>**: Octanol/Water partition coefficient .

**P<sub>c</sub>**: Critical Pressure (kPa).

**T<sub>boil</sub>**: Normal Boiling Point Temperature (K).

**T<sub>c</sub>**: Critical Temperature (K).

**T<sub>fus</sub>**: Normal melting (fusion) point (K).

**V<sub>c</sub>**: Critical Volume (m<sup>3</sup>/kg-mol).

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