

# Diethylmalonic acid, decyl 3-methylbenzyl ester

Inchi:	InChI=1S/C25H40O4/c1-5-8-9-10-11-12-13-14-18-28-23(26)25(6-2,7-3)24(27)29-20-22-1
InchiKey:	YUUMSYLWKIBWBV-UHFFFAOYSA-N
Formula:	C25H40O4
SMILES:	CCCCCCCCCOC(=O)C(CC)(CC)C(=O)OCc1cccc(C)c1
Mol. weight [g/mol]:	404.58

## Physical Properties

Property code	Value	Unit	Source
gf	-202.60	kJ/mol	Joback Method
hf	-832.62	kJ/mol	Joback Method
hfus	52.32	kJ/mol	Joback Method
hvap	91.20	kJ/mol	Joback Method
log10ws	-7.43		Crippen Method
logp	6.529		Crippen Method
mcvol	354.230	ml/mol	McGowan Method
pc	982.08	kPa	Joback Method
rinsol	2648.00		NIST Webbook
tb	952.41	K	Joback Method
tc	1166.68	K	Joback Method
tf	557.19	K	Joback Method
vc	1.365	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1181.34	J/molxK	952.41	Joback Method
cpg	1254.81	J/molxK	1130.97	Joback Method
cpg	1242.56	J/molxK	1095.26	Joback Method
cpg	1229.15	J/molxK	1059.55	Joback Method
cpg	1214.52	J/molxK	1023.83	Joback Method
cpg	1198.60	J/molxK	988.12	Joback Method
cpg	1265.98	J/molxK	1166.68	Joback Method
dvisc	0.0000203	Paxs	952.41	Joback Method
dvisc	0.0000269	Paxs	886.54	Joback Method

dvisc	0.0000373	Paxs	820.67	Joback Method
dvisc	0.0000548	Paxs	754.80	Joback Method
dvisc	0.0000868	Paxs	688.93	Joback Method
dvisc	0.0001513	Paxs	623.06	Joback Method
dvisc	0.0003009	Paxs	557.19	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369311&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369311&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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