

# Acetic acid, chloro, 1,1-dimethylpropyl ester

<b>Inchi:</b>	InChI=1S/C7H13ClO2/c1-4-7(2,3)10-6(9)5-8/h4-5H2,1-3H3
<b>InchiKey:</b>	KPFTZXFNGNJOQM-UHFFFAOYSA-N
<b>Formula:</b>	C7H13ClO2
<b>SMILES:</b>	CCC(C)(C)OC(=O)CCl
<b>Mol. weight [g/mol]:</b>	164.63

## Physical Properties

Property code	Value	Unit	Source
gf	-234.95	kJ/mol	Joback Method
hf	-457.10	kJ/mol	Joback Method
hfus	13.46	kJ/mol	Joback Method
hvap	43.42	kJ/mol	Joback Method
log10ws	-1.88		Crippen Method
logp	1.957		Crippen Method
mcvol	129.170	ml/mol	McGowan Method
pc	2884.30	kPa	Joback Method
rinpol	995.00		NIST Webbook
rinpol	995.00		NIST Webbook
ripol	1382.00		NIST Webbook
tb	470.05	K	Joback Method
tc	664.25	K	Joback Method
tf	273.15	K	Joback Method
vc	0.489	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	268.23	J/mol×K	470.05	Joback Method
cpg	321.60	J/mol×K	631.88	Joback Method
cpg	312.05	J/mol×K	599.52	Joback Method
cpg	301.96	J/mol×K	567.15	Joback Method
cpg	291.30	J/mol×K	534.78	Joback Method
cpg	280.06	J/mol×K	502.42	Joback Method
cpg	330.63	J/mol×K	664.25	Joback Method

dvisc	0.0002819	Paxs	470.05	Joback Method
dvisc	0.0003728	Paxs	437.23	Joback Method
dvisc	0.0005158	Paxs	404.42	Joback Method
dvisc	0.0007558	Paxs	371.60	Joback Method
dvisc	0.0011926	Paxs	338.78	Joback Method
dvisc	0.0020752	Paxs	305.97	Joback Method
dvisc	0.0041252	Paxs	273.15	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R115618&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R115618&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307i">http://pubs.acs.org/doi/abs/10.1021/ci990307i</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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