

# Spiro[8,8-dimethyl-1,2,3,4,5,6,7,8-octahydronaphthalene]

**Inchi:** InChI=1S/C17H26O/c1-16(2)9-5-6-13-8-11-17(12-14(13)16)10-4-3-7-15(17)18/h3-12H2,  
**InchiKey:** POCWDQYCFAMVQT-UHFFFAOYSA-N  
**Formula:** C17H26O  
**SMILES:** CC1(C)CCCC2=C1CC1(CCCCC1=O)CC2  
**Mol. weight [g/mol]:** 246.39

## Physical Properties

Property code	Value	Unit	Source
gf	86.75	kJ/mol	Joback Method
hf	-264.81	kJ/mol	Joback Method
hfus	7.88	kJ/mol	Joback Method
hvap	58.08	kJ/mol	Joback Method
log10ws	-5.27		Crippen Method
logp	4.807		Crippen Method
mcvol	215.080	ml/mol	McGowan Method
pc	2171.40	kPa	Joback Method
ripol	1929.40		NIST Webbook
ripol	1943.70		NIST Webbook
ripol	2427.60		NIST Webbook
ripol	2447.60		NIST Webbook
ripol	2427.60		NIST Webbook
tb	716.29	K	Joback Method
tc	976.75	K	Joback Method
tf	460.11	K	Joback Method
vc	0.800	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	658.76	J/molxK	716.29	Joback Method
cpg	683.80	J/molxK	759.70	Joback Method
cpg	707.91	J/molxK	803.11	Joback Method
cpg	731.47	J/molxK	846.52	Joback Method
cpg	754.86	J/molxK	889.93	Joback Method

cpg	778.47	J/mol×K	933.34	Joback Method
cpg	802.67	J/mol×K	976.75	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R299073&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R299073&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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