

# 1,2,3-Propanetricarboxylic acid, 2-hydroxy-, trihexyl ester

Inchi:	InChI=1S/C24H44O7/c1-4-7-10-13-16-29-21(25)19-24(28,23(27)31-18-15-12-9-6-3)20-2
InchiKey:	AMMPRZCMKXDUNE-UHFFFAOYSA-N
Formula:	C24H44O7
SMILES:	CCCCCCOC(=O)CC(O)(CC(=O)OCCCCC)C(=O)OCCCCC
Mol. weight [g/mol]:	444.60
CAS:	16544-70-0

## Physical Properties

Property code	Value	Unit	Source
gf	-684.54	kJ/mol	Joback Method
hf	-1434.07	kJ/mol	Joback Method
hfus	62.95	kJ/mol	Joback Method
hvap	111.87	kJ/mol	Joback Method
log10ws	-5.83		Crippen Method
logp	4.868		Crippen Method
mcvol	377.210	ml/mol	McGowan Method
pc	933.49	kPa	Joback Method
tb	1066.34	K	Joback Method
tc	1331.50	K	Joback Method
tf	639.96	K	Joback Method
vc	1.460	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1329.05	J/molxK	1066.34	Joback Method
cpg	1345.48	J/molxK	1110.53	Joback Method
cpg	1359.77	J/molxK	1154.73	Joback Method
cpg	1372.00	J/molxK	1198.92	Joback Method
cpg	1382.28	J/molxK	1243.12	Joback Method
cpg	1390.67	J/molxK	1287.31	Joback Method
cpg	1397.27	J/molxK	1331.50	Joback Method
dvisc	0.0000556	Paxs	639.96	Joback Method
dvisc	0.0000231	Paxs	711.02	Joback Method

dvisc	0.0000112	Paxs	782.09	Joback Method
dvisc	0.0000062	Paxs	853.15	Joback Method
dvisc	0.0000037	Paxs	924.21	Joback Method
dvisc	0.0000024	Paxs	995.28	Joback Method
dvisc	0.0000016	Paxs	1066.34	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C16544700&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C16544700&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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