

# 2-Hexene, (E)-

<b>Other names:</b>	(E)-2-C6H12 (E)-2-HEXENE TRANS-2-HEXENE trans-hex-2-ene
<b>Inchi:</b>	InChI=1S/C6H12/c1-3-5-6-4-2/h3,5H,4,6H2,1-2H3/b5-3+
<b>InchiKey:</b>	RYPKRALMXUUNKS-HWKANZROSA-N
<b>Formula:</b>	C6H12
<b>SMILES:</b>	CC=CCCC
<b>Mol. weight [g/mol]:</b>	84.16
<b>CAS:</b>	4050-45-7

## Physical Properties

Property code	Value	Unit	Source
af	0.2420		KDB
ap	299.150	K	KDB
gf	76.49	kJ/mol	KDB
hcg	3990.57	kJ/mol	KDB
hcn	3726.480	kJ/mol	KDB
hf	-52.10 ± 1.00	kJ/mol	NIST Webbook
hf	-52.50	kJ/mol	NIST Webbook
hf	-51.60 ± 0.80	kJ/mol	NIST Webbook
hf	-51.00 ± 0.80	kJ/mol	NIST Webbook
hf	-53.93	kJ/mol	KDB
hf	-55.80 ± 1.30	kJ/mol	NIST Webbook
hfl	-82.60 ± 0.80	kJ/mol	NIST Webbook
hfl	-87.40 ± 1.30	kJ/mol	NIST Webbook
hfl	-83.22 ± 0.84	kJ/mol	NIST Webbook
hfl	-83.70 ± 1.00	kJ/mol	NIST Webbook
hfus	11.50	kJ/mol	Joback Method
hvap	32.20	kJ/mol	NIST Webbook
hvap	31.50	kJ/mol	NIST Webbook
hvap	31.50	kJ/mol	NIST Webbook
hvap	31.60	kJ/mol	NIST Webbook
ie	8.97 ± 0.01	eV	NIST Webbook
ie	9.16 ± 0.01	eV	NIST Webbook
ie	8.93	eV	NIST Webbook
ie	8.97 ± 0.02	eV	NIST Webbook

log10ws	-2.19		Crippen Method
logp	2.363		Crippen Method
mcvol	91.100	ml/mol	McGowan Method
pc	3270.00	kPa	KDB
rinpol	605.50		NIST Webbook
rinpol	597.00		NIST Webbook
rinpol	596.60		NIST Webbook
rinpol	597.00		NIST Webbook
rinpol	597.00		NIST Webbook
rinpol	597.30		NIST Webbook
rinpol	597.00		NIST Webbook
rinpol	597.00		NIST Webbook
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rinpol	597.00		NIST Webbook
rinpol	598.00		NIST Webbook
rinpol	613.00		NIST Webbook
rinpol	596.00		NIST Webbook
rinpol	596.70		NIST Webbook
rinpol	606.00		NIST Webbook
rinpol	604.60		NIST Webbook
rinpol	606.00		NIST Webbook
rinpol	603.38		NIST Webbook
rinpol	603.43		NIST Webbook
rinpol	603.00		NIST Webbook
rinpol	604.00		NIST Webbook
rinpol	596.90		NIST Webbook
rinpol	597.00		NIST Webbook
rinpol	598.00		NIST Webbook
rinpol	613.00		NIST Webbook
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rinpol	596.60	NIST Webbook
rinpol	596.00	NIST Webbook
rinpol	606.00	NIST Webbook
ripol	663.40	NIST Webbook
ripol	663.40	NIST Webbook
ripol	663.00	NIST Webbook

ripol	648.00		NIST Webbook
ripol	660.00		NIST Webbook
ripol	663.40		NIST Webbook
ripol	663.40		NIST Webbook
ripol	663.00		NIST Webbook
ripol	663.00		NIST Webbook
ripol	663.40		NIST Webbook
tb	340.27 ± 0.40	K	NIST Webbook
tb	341.06 ± 0.30	K	NIST Webbook
tb	339.15 ± 2.00	K	NIST Webbook
tb	341.02 ± 0.30	K	NIST Webbook
tb	341.00	K	KDB
tb	340.24 ± 0.20	K	NIST Webbook
tb	341.10 ± 0.40	K	NIST Webbook
tb	341.06 ± 0.40	K	NIST Webbook
tb	341.10	K	NIST Webbook
tb	341.25 ± 0.50	K	NIST Webbook
tb	341.23 ± 0.30	K	NIST Webbook
tb	341.23 ± 0.50	K	NIST Webbook
tb	341.23 ± 0.40	K	NIST Webbook
tb	340.28 ± 0.30	K	NIST Webbook
tc	516.00	K	KDB
tc	509.00	K	Gas-Liquid Critical Temperatures of Some Alkenes, Amines, and Cyclic Hydrocarbons
tf	139.75 ± 0.60	K	NIST Webbook
tf	140.00	K	KDB
tf	140.17 ± 0.04	K	NIST Webbook
tf	140.08 ± 0.20	K	NIST Webbook
tf	139.75 ± 0.40	K	NIST Webbook
tf	139.65 ± 1.50	K	NIST Webbook
tf	139.99 ± 0.40	K	NIST Webbook
vc	0.351	m <sup>3</sup> /kmol	KDB
zc	0.2675280		KDB

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	200.78	J/mol×K	513.19	Joback Method
cpg	145.03	J/mol×K	340.84	Joback Method
cpg	155.37	J/mol×K	369.56	Joback Method

cpg	165.27	J/molxK	398.29	Joback Method
cpg	174.75	J/molxK	427.01	Joback Method
cpg	183.82	J/molxK	455.74	Joback Method
cpg	192.49	J/molxK	484.46	Joback Method
dvisc	0.0001858	Paxs	340.84	Joback Method
dvisc	0.0040974	Paxs	152.30	Joback Method
dvisc	0.0015745	Paxs	183.72	Joback Method
dvisc	0.0008000	Paxs	215.15	Joback Method
dvisc	0.0004831	Paxs	246.57	Joback Method
dvisc	0.0003269	Paxs	277.99	Joback Method
dvisc	0.0002395	Paxs	309.42	Joback Method
hvapt	28.91	kJ/mol	341.00	KDB
rfi	1.39073		298.15	KDB
rhol	678.00	kg/m3	293.00	KDB
srf	0.02	N/m	298.20	KDB

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.41268e+01
Coeff. B	-2.88478e+03
Coeff. C	-3.76300e+01
Temperature range (K), min.	246.08
Temperature range (K), max.	364.88

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C*\ln(T) + D*T^2$
Coeff. A	7.51774e+01
Coeff. B	-6.19923e+03
Coeff. C	-9.12514e+00
Coeff. D	7.24225e-06
Temperature range (K), min.	140.17
Temperature range (K), max.	513.00

# Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=198">https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=198</a>
<b>The Yaws Handbook of Vapor Pressure: Gas-Liquid Critical Temperatures of Some Alkenes, Amines, and Cyclic Hydrocarbons</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>KDB Vapor Pressure Data:</b>	<a href="https://www.doi.org/10.1021/je0341357">https://www.doi.org/10.1021/je0341357</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4050457&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4050457&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemo.com/doc/models/crippen_log10ws">https://www.chemo.com/doc/models/crippen_log10ws</a>

# Legend

<b>af:</b>	Acentric Factor
<b>ap:</b>	Aniline Point
<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>d<sub>visc</sub>:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>h<sub>cg</sub>:</b>	Heat of Combustion, Gross form
<b>h<sub>cn</sub>:</b>	Heat of Combustion, Net Form
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fl</sub>:</b>	Liquid phase enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>h<sub>vapt</sub>:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>p<sub>vap</sub>:</b>	Vapor pressure
<b>r<sub>fi</sub>:</b>	Refractive Index
<b>ρ<sub>ol</sub>:</b>	Liquid Density
<b>r<sub>inpol</sub>:</b>	Non-polar retention indices
<b>r<sub>ipol</sub>:</b>	Polar retention indices
<b>s<sub>rf</sub>:</b>	Surface Tension
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point  
**vc:** Critical Volume  
**zc:** Critical Compressibility

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