

val-ser

Inchi:

InChI=1S/C8H16N2O4/c1-4(2)6(9)7(12)10-5(3-11)8(13)14/h4-6,11H,3,9H2,1-2H3,(H,10,

InchiKey:

STTYIMSDIYISRG-UHFFFAOYSA-N

Formula:

C8H16N2O4

SMILES:

CC(C)C(N)C(=O)NC(CO)C(=O)O

Mol. weight [g/mol]:

204.22

CAS:

13588-94-8

Physical Properties

Property code	Value	Unit	Source
basg	874.10	kJ/mol	NIST Webbook
gf	-366.48	kJ/mol	Joback Method
hf	-666.65	kJ/mol	Joback Method
hfus	27.58	kJ/mol	Joback Method
hvap	96.17	kJ/mol	Joback Method
log10ws	0.09		Crippen Method
logp	-1.469		Crippen Method
mcvol	158.420	ml/mol	McGowan Method
pc	3995.65	kPa	Joback Method
tb	795.92	K	Joback Method
tc	988.60	K	Joback Method
tf	492.34	K	Joback Method
vc	0.580	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	476.06	J/mol×K	795.92	Joback Method
cpg	484.71	J/mol×K	828.03	Joback Method
cpg	492.79	J/mol×K	860.15	Joback Method
cpg	500.33	J/mol×K	892.26	Joback Method
cpg	507.34	J/mol×K	924.37	Joback Method
cpg	513.85	J/mol×K	956.49	Joback Method
cpg	519.88	J/mol×K	988.60	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C13588948&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

basg:	Gas basicity
cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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