

# Cyclohexane, 1,1,3,5-tetramethyl-, cis-

<b>Other names:</b>	1,1,3,5-Tetramethylcyclohexane, (Z)- cis-1,1,3,5-Tetramethylcyclohexane
<b>Inchi:</b>	InChI=1S/C10H20/c1-8-5-9(2)7-10(3,4)6-8/h8-9H,5-7H2,1-4H3/t8-,9+
<b>InchiKey:</b>	WOJSMJIXPQLESQ-DTORHVGOSA-N
<b>Formula:</b>	C10H20
<b>SMILES:</b>	CC1CC(C)CC(C)(C)C1
<b>Mol. weight [g/mol]:</b>	140.27
<b>CAS:</b>	50876-32-9

## Physical Properties

Property code	Value	Unit	Source
gf	36.86	kJ/mol	Joback Method
hf	-220.85	kJ/mol	Joback Method
hfus	9.33	kJ/mol	Joback Method
hvap	36.51	kJ/mol	Joback Method
log10ws	-3.18		Crippen Method
logp	3.469		Crippen Method
mcvol	140.900	ml/mol	McGowan Method
pc	2515.07	kPa	Joback Method
rinpol	908.00		NIST Webbook
rinpol	908.00		NIST Webbook
tb	425.60 ± 1.00	K	NIST Webbook
tb	425.11 ± 0.20	K	NIST Webbook
tb	425.70	K	NIST Webbook
tc	643.42	K	Joback Method
tf	225.26	K	Joback Method
vc	0.524	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	300.31	J/mol×K	438.65	Joback Method
cpg	321.00	J/mol×K	472.78	Joback Method
cpg	340.50	J/mol×K	506.91	Joback Method

cpg	358.88	J/mol×K	541.03	Joback Method
cpg	376.24	J/mol×K	575.16	Joback Method
cpg	392.65	J/mol×K	609.29	Joback Method
cpg	408.21	J/mol×K	643.42	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.43628e+01
Coeff. B	-3.56863e+03
Coeff. C	-5.94810e+01
Temperature range (K), min.	313.02
Temperature range (K), max.	453.74

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C50876329&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C50876329&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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