

# Benzenehexacarboxylic acid, hexamethyl ester

<b>Other names:</b>	Hexamethyl benzenehexacarboxylate Hexamethyl mellitate Mellitic acid, hexamethyl ester
<b>Inchi:</b>	InChI=1S/C18H18O12/c1-25-13(19)7-8(14(20)26-2)10(16(22)28-4)12(18(24)30-6)11(17(
<b>InchiKey:</b>	BQLICNRRYLYEDI-UHFFFAOYSA-N
<b>Formula:</b>	C18H18O12
<b>SMILES:</b>	<chem>COC(=O)c1c(C(=O)OC)c(C(=O)OC)c(C(=O)OC)c(C(=O)OC)c1C(=O)OC</chem>
<b>Mol. weight [g/mol]:</b>	426.33
<b>CAS:</b>	6237-59-8

## Physical Properties

Property code	Value	Unit	Source
chs	-7544.80 ± 1.70	kJ/mol	NIST Webbook
gf	-1238.58	kJ/mol	Joback Method
hf	-1704.47	kJ/mol	Joback Method
hfs	-2110.80 ± 1.90	kJ/mol	NIST Webbook
hfus	51.19	kJ/mol	Joback Method
hsub	154.30 ± 1.20	kJ/mol	NIST Webbook
hvap	116.18	kJ/mol	Joback Method
log10ws	-2.93		Crippen Method
logp	0.406		Crippen Method
mcvol	285.360	ml/mol	McGowan Method
pc	1716.03	kPa	Joback Method
rinpol	2535.00		NIST Webbook
tb	1120.56	K	Joback Method
tc	1372.35	K	Joback Method
tf	463.70 ± 0.20	K	NIST Webbook
vc	1.079	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	840.08	J/molxK	1330.38	Joback Method
cpg	823.47	J/molxK	1372.35	Joback Method

cpg	873.47	J/mol×K	1120.56	Joback Method
cpg	873.00	J/mol×K	1162.52	Joback Method
cpg	869.55	J/mol×K	1204.49	Joback Method
cpg	863.00	J/mol×K	1246.45	Joback Method
cpg	853.21	J/mol×K	1288.42	Joback Method
dvisc	0.0000196	Paxs	1120.56	Joback Method
dvisc	0.0000232	Paxs	1069.57	Joback Method
dvisc	0.0000745	Paxs	814.60	Joback Method
dvisc	0.0000559	Paxs	865.59	Joback Method
dvisc	0.0000432	Paxs	916.59	Joback Method
dvisc	0.0000344	Paxs	967.58	Joback Method
dvisc	0.0000279	Paxs	1018.57	Joback Method
hfust	22.50	kJ/mol	463.70	NIST Webbook
hsubt	140.70 ± 1.10	kJ/mol	412.50	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6237598&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6237598&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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