

# Benzene, 1-iodo-3-(trifluoromethyl)-

<b>Other names:</b>	m-Iodobenzotrifluoride 3-Iodobenzotrifluoride 3-(Trifluoromethyl)iodobenzene «alpha», «alpha», «alpha»-Trifluoro-m-iodotoluene m-(Trifluoromethyl)iodobenzene Toluene, «alpha», «alpha», «alpha»-trifluoro-m-iodo-1-iodo-3-(trifluoromethyl)benzene m-Iodo-«alpha», «alpha», «alpha»-trifluorotoluene
<b>Inchi:</b>	InChI=1S/C7H4F3I/c8-7(9,10)5-2-1-3-6(11)4-5/h1-4H
<b>InchiKey:</b>	IGISPMBUGPHLBY-UHFFFAOYSA-N
<b>Formula:</b>	C7H4F3I
<b>SMILES:</b>	FC(F)(F)c1cccc(I)c1
<b>Mol. weight [g/mol]:</b>	272.01
<b>CAS:</b>	401-81-0

## Physical Properties

Property code	Value	Unit	Source
gf	-412.63	kJ/mol	Joback Method
hf	-482.96	kJ/mol	Joback Method
hfus	13.77	kJ/mol	Joback Method
hvap	39.74	kJ/mol	Joback Method
log10ws	-3.72		Crippen Method
logp	3.310		Crippen Method
mcvol	116.860	ml/mol	McGowan Method
pc	3411.87	kPa	Joback Method
rinpol	1015.60		NIST Webbook
rinpol	1023.00		NIST Webbook
rinpol	1015.60		NIST Webbook
tb	478.94	K	Joback Method
tc	707.75	K	Joback Method
tf	269.84	K	Joback Method
vc	0.451	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	211.40	J/mol×K	478.94	Joback Method
cpg	221.30	J/mol×K	517.08	Joback Method
cpg	230.28	J/mol×K	555.21	Joback Method
cpg	238.43	J/mol×K	593.35	Joback Method
cpg	245.80	J/mol×K	631.48	Joback Method
cpg	252.48	J/mol×K	669.62	Joback Method
cpg	258.51	J/mol×K	707.75	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	355.50	K	3.30	NIST Webbook
tbrp	355.00	K	3.30	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C401810&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C401810&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l

<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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