

# Benzene, pentafluoromethoxy-

<b>Other names:</b>	Pentafluoroanisole 2,3,4,5,6-Pentafluoroanisole Anisole, 2,3,4,5,6-pentafluoro- Methyl pentafluorophenyl ether Pentafluorophenyl methyl ether
<b>Inchi:</b>	InChI=1S/C7H3F5O/c1-13-7-5(11)3(9)2(8)4(10)6(7)12/h1H3
<b>InchiKey:</b>	ZRQUIRABLIQJRI-UHFFFAOYSA-N
<b>Formula:</b>	C7H3F5O
<b>SMILES:</b>	COc1c(F)c(F)c(F)c(F)c1F
<b>Mol. weight [g/mol]:</b>	198.09
<b>CAS:</b>	389-40-2

## Physical Properties

Property code	Value	Unit	Source
ea	0.54 ± 0.09	eV	NIST Webbook
gf	-1006.73	kJ/mol	Joback Method
hf	-1121.40	kJ/mol	Joback Method
hfus	22.57	kJ/mol	Joback Method
hvap	35.09	kJ/mol	Joback Method
ie	9.10 ± 0.02	eV	NIST Webbook
log10ws	-3.25		Crippen Method
logp	2.391		Crippen Method
mvol	100.450	ml/mol	McGowan Method
pc	2841.41	kPa	Joback Method
tb	411.00	K	NIST Webbook
tb	411.70	K	NIST Webbook
tc	596.67	K	Joback Method
tf	282.85	K	Joback Method
vc	0.427	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	205.28	J/mol×K	429.91	Joback Method

cpg	211.81	J/mol×K	457.70	Joback Method
cpg	218.20	J/mol×K	485.50	Joback Method
cpg	224.42	J/mol×K	513.29	Joback Method
cpg	230.49	J/mol×K	541.08	Joback Method
cpg	236.38	J/mol×K	568.87	Joback Method
cpg	242.09	J/mol×K	596.67	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C389402&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C389402&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>ea:</b>	Electron affinity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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