

# Thiamylal

<b>Other names:</b>	4,6(1H,5H)-Pyrimidinedione, dihydro-5-(1-methylbutyl)-5-(2-propenyl)-2-thioxo-5-Allyl-5-(1-methylbutyl)-2-thiobarbituric acid Barbituric acid, 5-allyl-5-(1-methylbutyl)-2-thio- Surital Thioseconal
<b>Inchi:</b>	InChI=1S/C12H18N2O2S/c1-4-6-8(3)12(7-5-2)9(15)13-11(17)14-10(12)16/h5,8H,2,4,6-7
<b>InchiKey:</b>	XLOMZPUITCYLMJ-UHFFFAOYSA-N
<b>Formula:</b>	C12H18N2O2S
<b>SMILES:</b>	C=CCC1(C(C)CCC)C(=O)NC(=S)NC1=O
<b>Mol. weight [g/mol]:</b>	254.35
<b>CAS:</b>	77-27-0

## Physical Properties

Property code	Value	Unit	Source
gf	175.61	kJ/mol	Joback Method
hf	-185.98	kJ/mol	Joback Method
hfus	31.80	kJ/mol	Joback Method
hvap	70.01	kJ/mol	Joback Method
log10ws	-3.46		Aqueous Solubility Prediction Method
log10ws	-3.46		Estimated Solubility Method
logp	1.516		Crippen Method
mcvol	199.930	ml/mol	McGowan Method
pc	2811.36	kPa	Joback Method
rinpol	1873.00		NIST Webbook
rinpol	1899.00		NIST Webbook
rinpol	1873.00		NIST Webbook
rinpol	1886.00		NIST Webbook
rinpol	1886.00		NIST Webbook
tb	795.37	K	Joback Method
tc	1050.18	K	Joback Method
tf	405.65	K	Aqueous Solubility Prediction Method
vc	0.740	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	577.95	J/molxK	795.37	Joback Method
cpg	595.16	J/molxK	837.84	Joback Method
cpg	611.67	J/molxK	880.31	Joback Method
cpg	627.58	J/molxK	922.77	Joback Method
cpg	643.00	J/molxK	965.24	Joback Method
cpg	658.03	J/molxK	1007.71	Joback Method
cpg	672.76	J/molxK	1050.18	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>Aqueous Solubility Prediction Method:</b>	<a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx</a>
<b>Estimated Solubility Method:</b>	<a href="http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt">http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C77270&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C77270&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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