

# 3-Hexen-2-one, 3,4-dimethyl-

|                      |                                                          |
|----------------------|----------------------------------------------------------|
| Other names:         | 3,4-Dimethyl-3-hexen-2-one<br>3,4-dimethylhex-3-en-2-one |
| Inchi:               | InChI=1S/C8H14O/c1-5-6(2)7(3)8(4)9/h5H2,1-4H3/b7-6+      |
| InchiKey:            | WRHRFVOAEDXVPC-VOTSOKGWSA-N                              |
| Formula:             | C8H14O                                                   |
| SMILES:              | CCC(C)=C(C)C(C)=O                                        |
| Mol. weight [g/mol]: | 126.20                                                   |
| CAS:                 | 1635-02-5                                                |

## Physical Properties

| Property code | Value   | Unit    | Source         |
|---------------|---------|---------|----------------|
| gf            | -49.32  | kJ/mol  | Joback Method  |
| hf            | -223.39 | kJ/mol  | Joback Method  |
| hfus          | 15.66   | kJ/mol  | Joback Method  |
| hvap          | 40.27   | kJ/mol  | Joback Method  |
| log10ws       | -2.30   |         | Crippen Method |
| logp          | 2.322   |         | Crippen Method |
| mcvol         | 120.850 | ml/mol  | McGowan Method |
| pc            | 2887.40 | kPa     | Joback Method  |
| tb            | 440.23  | K       | Joback Method  |
| tc            | 631.40  | K       | Joback Method  |
| tf            | 196.85  | K       | Joback Method  |
| vc            | 0.471   | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 239.59 | J/mol×K | 440.23          | Joback Method |
| cpg           | 252.19 | J/mol×K | 472.09          | Joback Method |
| cpg           | 264.18 | J/mol×K | 503.95          | Joback Method |
| cpg           | 275.57 | J/mol×K | 535.81          | Joback Method |
| cpg           | 286.40 | J/mol×K | 567.68          | Joback Method |
| cpg           | 296.69 | J/mol×K | 599.54          | Joback Method |
| cpg           | 306.46 | J/mol×K | 631.40          | Joback Method |

# Pressure Dependent Properties

| Property code | Value  | Unit | Pressure [kPa] | Source       |
|---------------|--------|------|----------------|--------------|
| tbrp          | 338.20 | K    | 2.70           | NIST Webbook |

## Correlations

| Information                 | Value                         |
|-----------------------------|-------------------------------|
| Property code               | pvap                          |
| Equation                    | $\ln(P_{vp}) = A + B/(T + C)$ |
| Coeff. A                    | 1.25428e+01                   |
| Coeff. B                    | -3.17939e+03                  |
| Coeff. C                    | -6.12880e+01                  |
| Temperature range (K), min. | 320.72                        |
| Temperature range (K), max. | 500.96                        |

## Sources

|                                      |                                                                                                                                                                                         |
|--------------------------------------|-----------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| Crippen Method:                      | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>                                                                               |
| Crippen Method:                      | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                                                                       |
| Joback Method:                       | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                                                                   |
| McGowan Method:                      | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                                                                   |
| NIST Webbook:                        | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1635025&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1635025&amp;Units=SI</a>                                             |
| The Yaws Handbook of Vapor Pressure: | <a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a> |

## Legend

|       |                                              |
|-------|----------------------------------------------|
| cpg:  | Ideal gas heat capacity                      |
| gf:   | Standard Gibbs free energy of formation      |
| hf:   | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions    |

|                            |                                                 |
|----------------------------|-------------------------------------------------|
| <b>h<sub>vap</sub>:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log<sub>10</sub>ws:</b> | Log10 of Water solubility in mol/l              |
| <b>log<sub>p</sub>:</b>    | Octanol/Water partition coefficient             |
| <b>mc<sub>vol</sub>:</b>   | McGowan's characteristic volume                 |
| <b>p<sub>c</sub>:</b>      | Critical Pressure                               |
| <b>p<sub>vap</sub>:</b>    | Vapor pressure                                  |
| <b>t<sub>b</sub>:</b>      | Normal Boiling Point Temperature                |
| <b>t<sub>brp</sub>:</b>    | Boiling point at reduced pressure               |
| <b>t<sub>c</sub>:</b>      | Critical Temperature                            |
| <b>t<sub>f</sub>:</b>      | Normal melting (fusion) point                   |
| <b>v<sub>c</sub>:</b>      | Critical Volume                                 |

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