

# Ethylphenylhydantoin

**Other names:**

2,4-Imidazolidinedione, 5-ethyl-5-phenyl-  
5-Ethyl-5-phenyl-2,4-imidazolidinedione  
5-Ethyl-5-phenylhydantoin  
5-Phenyl-5-ethylhydantoin  
5-ethyl-5-phenylimidazolidine-2,4-dione  
Hydantoin, 5-ethyl-5-phenyl-  
NSC 150466  
Nirvanol  
Normephenytoin

**Inchi:**

InChI=1S/C11H12N2O2/c1-2-11(8-6-4-3-5-7-8)9(14)12-10(15)13-11/h3-7H,2H2,1H3,(H2

**InchiKey:**

UDTWZFFJEMMUFLC-UHFFFAOYSA-N

**Formula:**

C11H12N2O2

**SMILES:**

CCC1(c2ccccc2)NC(=O)NC1=O

**Mol. weight [g/mol]:**

204.23

**CAS:**

631-07-2

## Physical Properties

Property code	Value	Unit	Source
gf	115.45	kJ/mol	Joback Method
hf	-157.90	kJ/mol	Joback Method
hfus	24.12	kJ/mol	Joback Method
hvap	63.47	kJ/mol	Joback Method
log10ws	-2.64		Aqueous Solubility Prediction Method
logp	1.131		Crippen Method
mcvol	154.330	ml/mol	McGowan Method
pc	3916.03	kPa	Joback Method
rinpol	1810.00		NIST Webbook
rinpol	1825.00		NIST Webbook
rinpol	1795.00		NIST Webbook
rinpol	1825.00		NIST Webbook
rinpol	1790.00		NIST Webbook
rinpol	1795.00		NIST Webbook
tb	726.02	K	Joback Method
tc	1004.96	K	Joback Method
tf	621.45	K	Joback Method
vc	0.571	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	430.06	J/mol×K	726.02	Joback Method
cpg	447.16	J/mol×K	772.51	Joback Method
cpg	463.33	J/mol×K	819.00	Joback Method
cpg	478.69	J/mol×K	865.49	Joback Method
cpg	493.36	J/mol×K	911.98	Joback Method
cpg	507.45	J/mol×K	958.47	Joback Method
cpg	521.07	J/mol×K	1004.96	Joback Method

## Sources

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**Aqueous Solubility Prediction Method:**

<http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa>

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C631072&Units=SI>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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