

# 9-Dodecyn-1-ol

<b>Inchi:</b>	InChI=1S/C12H22O/c1-2-3-4-5-6-7-8-9-10-11-12-13/h13H,2,5-12H2,1H3
<b>InchiKey:</b>	MIDXCKJJIFQOLT-UHFFFAOYSA-N
<b>Formula:</b>	C12H22O
<b>SMILES:</b>	CCC#CCCCCCCCCO
<b>Mol. weight [g/mol]:</b>	182.30
<b>CAS:</b>	71084-08-7

## Physical Properties

Property code	Value	Unit	Source
gf	116.14	kJ/mol	Joback Method
hf	-170.94	kJ/mol	Joback Method
hfus	34.05	kJ/mol	Joback Method
hvap	61.14	kJ/mol	Joback Method
log10ws	-3.90		Crippen Method
logp	3.123		Crippen Method
mcvol	177.210	ml/mol	McGowan Method
pc	2229.20	kPa	Joback Method
rinpol	1543.00		NIST Webbook
rinpol	1543.00		NIST Webbook
tb	575.14	K	Joback Method
tc	748.57	K	Joback Method
tf	391.92	K	Joback Method
vc	0.689	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	441.34	J/molxK	575.14	Joback Method
cpg	455.35	J/molxK	604.05	Joback Method
cpg	468.78	J/molxK	632.95	Joback Method
cpg	481.64	J/molxK	661.86	Joback Method
cpg	493.95	J/molxK	690.76	Joback Method
cpg	505.73	J/molxK	719.67	Joback Method
cpg	517.00	J/molxK	748.57	Joback Method

# Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.63353e+01
Coeff. B	-5.41743e+03
Coeff. C	-9.65180e+01
Temperature range (K), min.	434.15
Temperature range (K), max.	588.15

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C71084087&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C71084087&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/48-805-5/9-Dodecyn-1-ol.pdf>

Generated by Cheméo on 2024-04-25 21:23:26.129995874 +0000 UTC m=+16369455.050573186.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.