

# Fumaric acid, isohexyl 2-propylphenyl ester

<b>Inchi:</b>	InChI=1S/C19H26O4/c1-4-8-16-10-5-6-11-17(16)23-19(21)13-12-18(20)22-14-7-9-15(2)3
<b>InchiKey:</b>	XIZVTGZKYMHQEY-OUKQBFOZSA-N
<b>Formula:</b>	C19H26O4
<b>SMILES:</b>	CCCCc1ccccc1OC(=O)C=CC(=O)OCCCC(C)C
<b>Mol. weight [g/mol]:</b>	318.41

## Physical Properties

Property code	Value	Unit	Source
gf	-178.18	kJ/mol	Joback Method
hf	-588.09	kJ/mol	Joback Method
hfus	40.87	kJ/mol	Joback Method
hvap	78.71	kJ/mol	Joback Method
log10ws	-4.83		Crippen Method
logp	4.080		Crippen Method
mvol	265.390	ml/mol	McGowan Method
pc	1509.33	kPa	Joback Method
rinpol	2211.00		NIST Webbook
rinpol	2211.00		NIST Webbook
tb	822.08	K	Joback Method
tc	1028.56	K	Joback Method
tf	467.07	K	Joback Method
vc	1.014	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	793.60	J/molxK	822.08	Joback Method
cpg	809.20	J/molxK	856.49	Joback Method
cpg	823.72	J/molxK	890.91	Joback Method
cpg	837.19	J/molxK	925.32	Joback Method
cpg	849.65	J/molxK	959.74	Joback Method
cpg	861.13	J/molxK	994.15	Joback Method
cpg	871.66	J/molxK	1028.56	Joback Method
dvisc	0.0006893	Paxs	467.07	Joback Method

dvisc	0.0003486	Paxs	526.24	Joback Method
dvisc	0.0002023	Paxs	585.41	Joback Method
dvisc	0.0001298	Paxs	644.57	Joback Method
dvisc	0.0000897	Paxs	703.74	Joback Method
dvisc	0.0000656	Paxs	762.91	Joback Method
dvisc	0.0000503	Paxs	822.08	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348128&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348128&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/48-857-8/Fumaric-acid-isohehexyl-2-propylphenyl-ester.pdf>

Generated by Cheméo on 2024-04-27 09:30:28.717608967 +0000 UTC m=+16499477.638186284.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.