

# Cyclopentylcyclohexane

<b>Other names:</b>	Cyclohexane, cyclopentyl- Cyclohexylcyclopentane
<b>Inchi:</b>	InChI=1S/C11H20/c1-2-6-10(7-3-1)11-8-4-5-9-11/h10-11H,1-9H2
<b>InchiKey:</b>	WVABZRMBCQFXBK-UHFFFAOYSA-N
<b>Formula:</b>	C11H20
<b>SMILES:</b>	C1CCC(C2CCCC2)CC1
<b>Mol. weight [g/mol]:</b>	152.28
<b>CAS:</b>	1606-08-2

## Physical Properties

Property code	Value	Unit	Source
chl	-6956.70	kJ/mol	NIST Webbook
gf	102.74	kJ/mol	Joback Method
hf	-155.57	kJ/mol	Joback Method
hfus	10.02	kJ/mol	Joback Method
hvap	40.77	kJ/mol	Joback Method
log10ws	-3.73		Crippen Method
logp	3.757		Crippen Method
mcvol	144.130	ml/mol	McGowan Method
pc	2802.44	kPa	Joback Method
rinpol	1225.00		NIST Webbook
rinpol	1210.00		NIST Webbook
tb	489.00 ± 3.00	K	NIST Webbook
tb	488.30	K	NIST Webbook
tc	713.20	K	Joback Method
tf	232.01	K	Joback Method
vc	0.525	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	458.77	J/mol×K	713.20	Joback Method
cpg	441.37	J/mol×K	675.32	Joback Method
cpg	422.69	J/mol×K	637.44	Joback Method

cpg	402.70	J/mol×K	599.56	Joback Method
cpg	381.33	J/mol×K	561.67	Joback Method
cpg	358.53	J/mol×K	523.79	Joback Method
cpg	334.24	J/mol×K	485.91	Joback Method
dvisc	0.0074291	Paxs	232.01	Joback Method
dvisc	0.0003310	Paxs	485.91	Joback Method
dvisc	0.0004341	Paxs	443.59	Joback Method
dvisc	0.0006029	Paxs	401.28	Joback Method
dvisc	0.0009046	Paxs	358.96	Joback Method
dvisc	0.0015129	Paxs	316.64	Joback Method
dvisc	0.0029653	Paxs	274.33	Joback Method
hvapt	47.90	kJ/mol	435.50	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.21994e+01
Coeff. B	-2.53409e+03
Coeff. C	-1.48676e+02
Temperature range (K), min.	361.41
Temperature range (K), max.	516.58

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1606082&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1606082&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

**chl:** Standard liquid enthalpy of combustion

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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